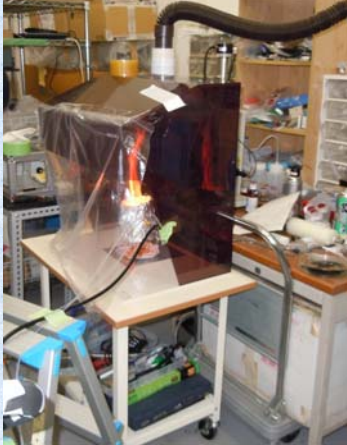


# Laboratory measurements of emission factors of nonmethane volatile organic compounds from burning of Chinese crop residues

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## Key Points

- EFs of VOCs from Chinese crop residue burning were measured in laboratory experiments
- EFs from the burning of both dry and wet samples were determined
- Residual moisture in crop residues suppressed VOC emissions from the burning

## Abstract

The emission factors (EFs) of non-methane volatile organic compounds (NMVOCs) emitted during the burning of Chinese crop residue were investigated as a function of modified combustion efficiency (MCE) in laboratory experiments. NMVOCs, including acetonitrile, aldehydes/ketones, furan, and aromatic hydrocarbons, were monitored by proton-transfer-reaction mass spectrometry (PTR-MS). Rape plant was burned in dry conditions and wheat straw was burned in both wet and dry conditions to simulate the possible burning of damp crop residue in regions of high temperature and humidity. We compared the present data to field data reported by *Kudo et al.* [2014]. Good agreement between field and laboratory data was obtained for aromatics under relatively more smoldering combustion of dry samples, but laboratory data were slightly overestimated compared to field data for oxygenated VOC (OVOC). When EFs from the burning of wet samples were investigated, the consistency between the field and laboratory data for OVOCs was stronger than for dry samples. This may be caused by residual moisture in crop residue that has been stockpiled in humid regions. Comparison of the wet laboratory data with field data suggests that *Kudo et al.* [2014] observed the biomass burning plumes under relatively more smoldering conditions in which approximately a few tens of percentages of burned fuel materials were wet.

## Introduction

### Biomass Burning (BB)

- ✓ The largest source of primary carbonaceous particles  
... affecting visibility, global climate, and human health
- ✓ The second largest source of trace gases  
NMVOCs → photochemical O<sub>3</sub>, SOA formation

### Central East China (CEC)

- ✓ Highly polluted area in Asia
- ✓ Large-scale burning of crop residues  
Wheat straw burning in May/June  
Rice straw burning in October

### Field Measurement of NMVOCs

- Location: Rudong town  
... rural site, 100 km north of Shanghai
- Period: June 2010
- BB plumes were often measured  
... grey areas in Figure 2
- Typical crop residues:

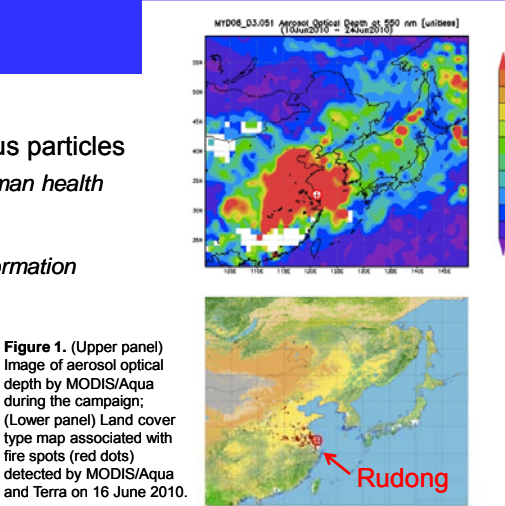
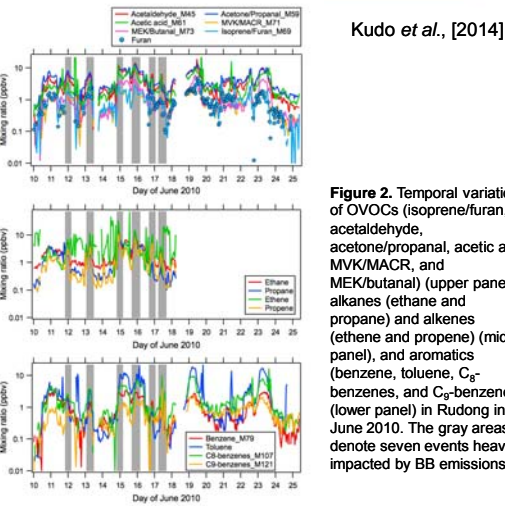


Figure 1. (Upper panel) Image of aerosol optical depth by MODIS/Aqua during the campaign; (Lower panel) Land cover type map associated with fire spots (red dots) detected by MODIS/Aqua and Terra on 16 June 2010.

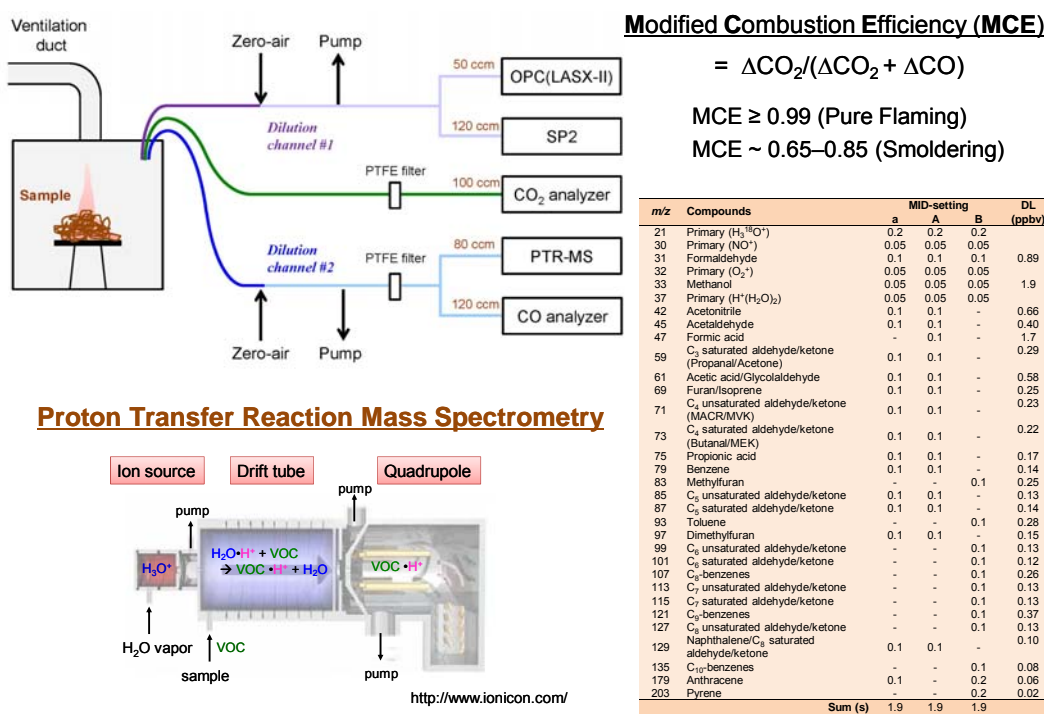


Kudo et al., [2014]

## Objectives

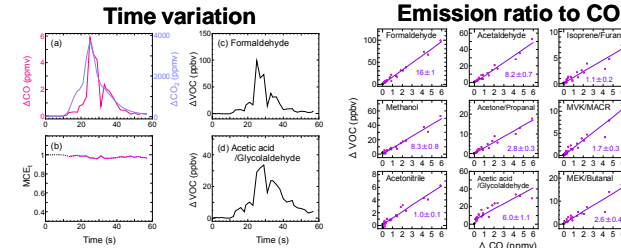
- Determination of EFs of NMVOCs from the BB of Chinese crop residues in Laboratory Experiments
- Comparison of EFs: Laboratory Experiments vs. Field Measurements

## Experimental Setup



## Results and Discussion

### Flaming-dominant combustion



### Relatively smoldering combustion

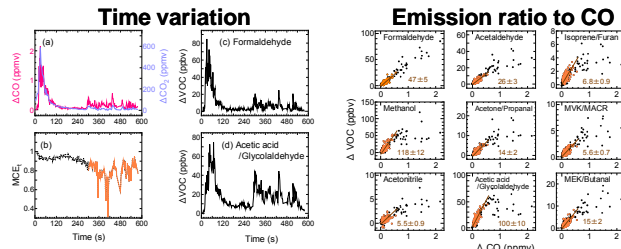
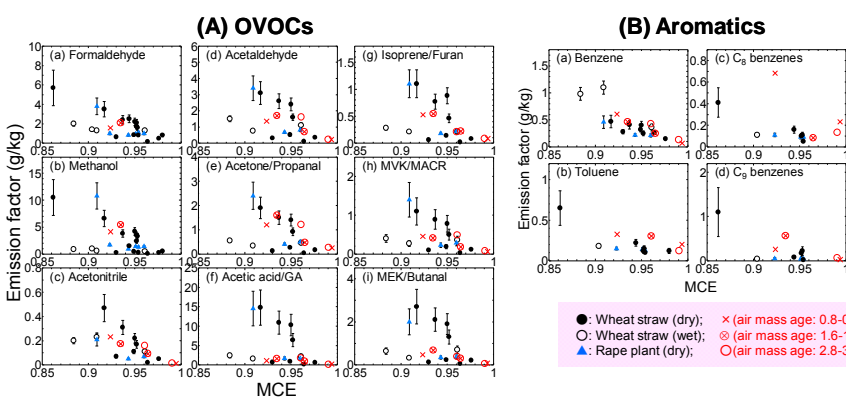


Table 1. Emission factors (g/kg) of NMVOCs from crop residue burning<sup>a</sup>

	Laboratory experiment <sup>b</sup>				Field measurement <sup>c</sup>			
	Wheat straw	Rape plant	Wheat straw	Rape plant	Wheat straw	Rape plant	Wheat straw	Rape plant
MCE	0.97(0.01)	0.93(0.07)	0.91(0.05)	0.92	0.94(0.03)	0.992	0.930	0.925
Formaldehyde	0.6(0.5)	2.7(3.0)	1.5(0.5)	0.9(0.1)	1.7(2.1)	-	1.87	1.93(1.32)
Methanol	0.3(0.3)	4.3(6.0)	0.7(0.3)	1.3(0.4)	3.7(7.1)	-	2.94	1.87(1.53)
Acetonitrile	0.06(0.05)	0.29(0.16)	0.18(0.07)	0.085	0.14(0.07)	0.01(0.00)	0.29(0.03)	0.225(0.173)
Acetaldehyde	0.3(0.2)	2.4(0.8)	1.1(0.4)	0.6(7)	2.1(1.3)	1.02	0.0067(0.0023)	2.68(2.42)
Acetone/Propanal	0.2(0.1)	1.4(0.5)	0.5(0.1)	0.4(1)	1.5(1.0)	0.24	0.83	0.11(0.02)
Acetic acid/Glycolaldehyde	0.6(0.4)	10.6(4.2)	2.1(0.5)	1(7)	8.1(6.4)	0.11 <sup>h</sup>	0.54 <sup>h</sup>	3.88(3.64)
Isoprene/Furan	0.08(0.07)	0.8(0.3)	0.23(0.05)	0.18	0.7(0.4)	0.08	0.52(0.01)	0.28(0.02) <sup>g</sup>
MVK/MACR	0.11(0.08)	0.8(0.3)	0.34(0.07)	0.23	0.9(0.6)	0.06	0.43(0.02)	0.607(0.515)
MEK/Butanol	0.16(0.10)	2.0(0.7)	0.6(0.2)	0.34	1.2(0.8)	0.05	0.28	0.0014(0.0004)
Benzene	0.25(0.10)	0.4(0.1)	0.8(0.5)	0.22	0.3(0.1)	0.09(0.02)	0.53(0.07)	0.31(0.08)
Toluene	0.11(0.01)	0.3(0.3)	0.18	0.15	0.14(0.01)	0.16(0.04)	0.32	0.14(0.06)
C <sub>8</sub> -benzenes	0.05	0.2(0.2)	0.11	0.11	0.09(0.00)	0.18(0.05)	0.68	0.09(0.05)
C <sub>9</sub> -benzenes	0.02	0.4(0.7)	0.04	0.05	0.04(0.00)	0.03(0.02)	0.40(0.15)	0.06(0.04)

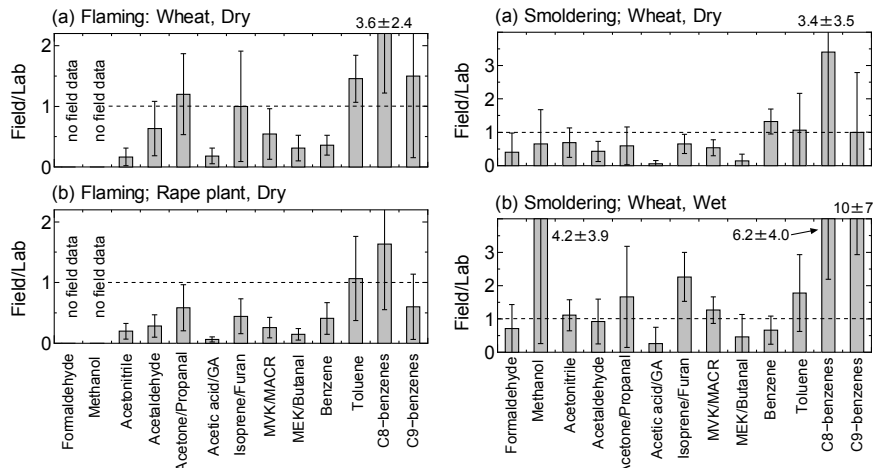
<sup>a</sup>The values in parentheses represent the ranges of the data (± error). The values in curly brackets are based on single data. <sup>b</sup>This work. <sup>c</sup>Kudo et al. [2014]. <sup>d</sup>Li et al. [2009]. <sup>e</sup>Zhang et al. [2013]. <sup>f</sup>Stockwell et al. [2015]. <sup>g</sup>Hatch et al. [2015]. <sup>h</sup>The EFs may be underestimated due to the condensation of water vapor in the sampling tube. <sup>i</sup>The EFs are determined by extrapolation from the measured normalized excess mixing ratios (NEMRs) and photochemical ages to the values at an air mass age of 0. The MCE was estimated from the reported EFs of CO<sub>2</sub> and CO. It was reported that the burning cycles were dominated by the flaming pattern. <sup>j</sup>The value was derived from the EF of total NMHCs and the fraction of each compound.

### Emission factors (EFs) of NMVOCs



- A trend that the EFs increase as the MCE decreases was observed.
- There is no meaningful difference between the wheat straw (dry) and the rape plant (dry) samples.
- The EFs of wet wheat straw are low relative to those of dry wheat straw for OVOCs.

### Laboratory Experiments vs. Field Measurements



- ✓ For flaming-dominant data, the field data for most OVOCs were smaller than the laboratory data.
- ✓ For relatively smoldering data, the field data were smaller than the laboratory data for OVOCs.
- ✓ The underestimations for OVOCs were improved with wet samples.

Yokelson et al., [1999]

$$EF_X \text{ (g/kg)} = F_C \times 1000 \text{ (g/kg)} \times MM_X \text{ (g)/} MM_C \text{ (g)} \times C_X / C_{\text{total}}$$

$F_C$ : the mass fraction of carbon in the fuel (assumed to be 0.5)  
 $MM_X$ : the molecular mass of compound X  
 $MM_C$ : the molecular mass of carbon (12.011 g mol<sup>-1</sup>)  
 $C_X / C_{\text{total}}$ : the number of emitted moles of compound X divided by the total number of moles of carbon emitted.

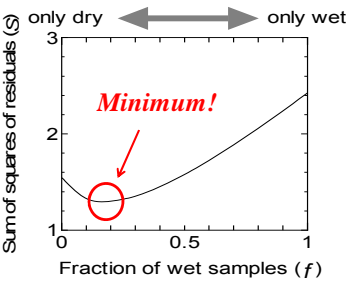
$$C_X / C_{\text{total}} = \Delta C_X / \Delta CO_2 \sum (n_{C_Y} \times \Delta C_Y / \Delta CO_2)$$

$\Delta C_X / \Delta CO_2$ : the fire-averaged emission ratio (ER) of species X to CO<sub>2</sub>  
 $n_{C_Y}$ : the number of carbon atoms in compound Y  
The sum is over all carbon-containing species including CO<sub>2</sub>, CO and CH<sub>4</sub>.

### Estimation of the fraction of wet sample in the field data

$$S = \sum_i (\log_{10} R_{F/L}(i))^2$$

$$R_{F/L}(i) = (1-f) \times R_{F/L}^d(i) + f \times R_{F/L}^w(i)$$



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