

Trace gases and aerosols in the AMA as observed by airborne infrared limb-imaging during StratoClim, Kathmandu, July 2017

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Institute of Meteorology and Climate Research



Two nobel prizes 100 years ago

Chemistry

Fritz Haber

The Nobel Prize in Chemistry 1918 was received by Fritz Haber in 1919: "for the synthesis of ammonia from its elements."

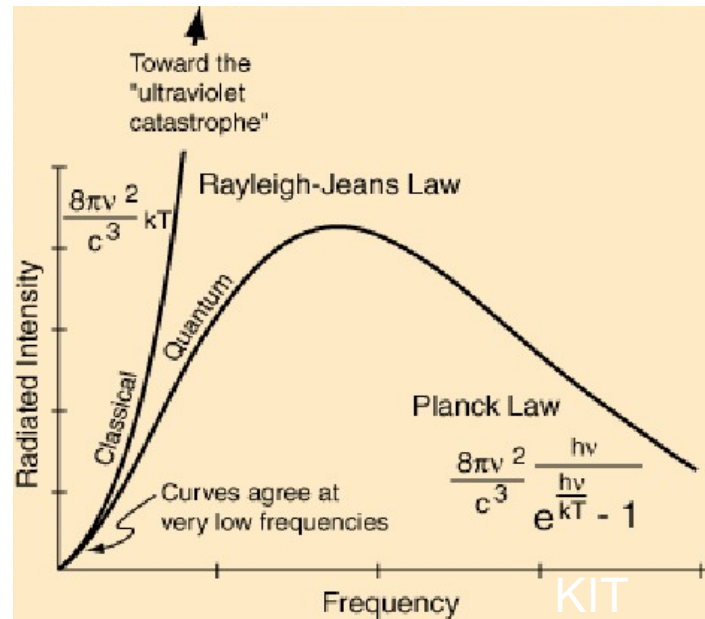


Nobel foundation archive

Physics

Max Planck

The Nobel Prize in Physics 1918 was received by Max Planck in 1919: "in recognition of the services he rendered to the advancement of Physics by his discovery of energy quanta."

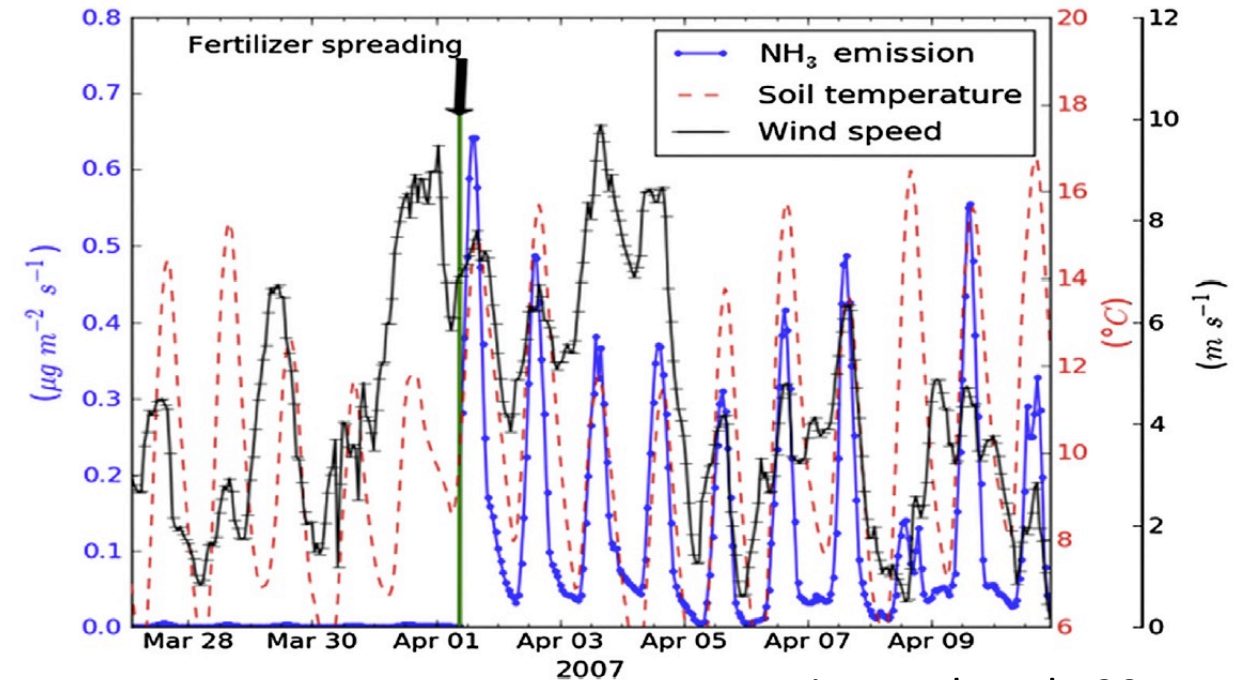


Nobel foundation archive

Significance of ammonia (NH₃)

- Main alkaline species in the atmosphere
- Major source: **agriculture**
- Formation of aerosols by neutralization of acids: **ammonium sulfate** and **ammonium nitrate** depending on the availability of H₂SO₄ and HNO₃
- Important fraction of fine particulate matter
- **Increase of NH₃ emissions in the future:** compensation of aerosol radiative forcing change by reduction of SO₂ emissions
- Important for the **initial nucleation of sulfate aerosols under cold temperatures**

But: Difficult to measure in-situ



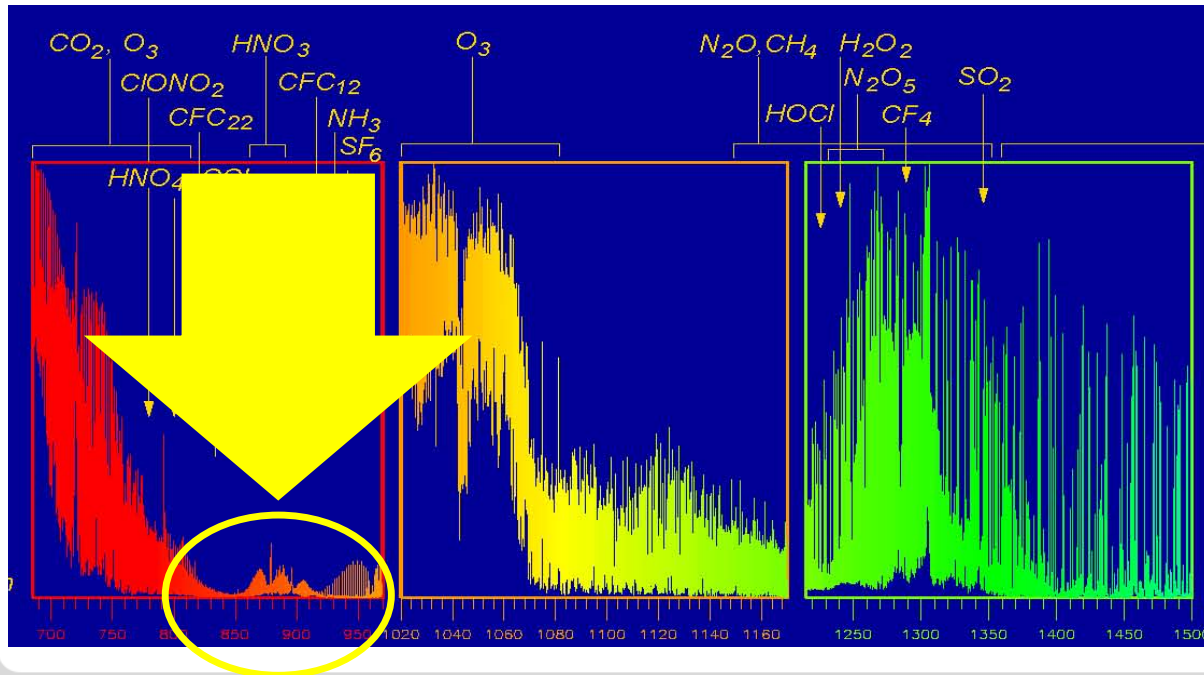
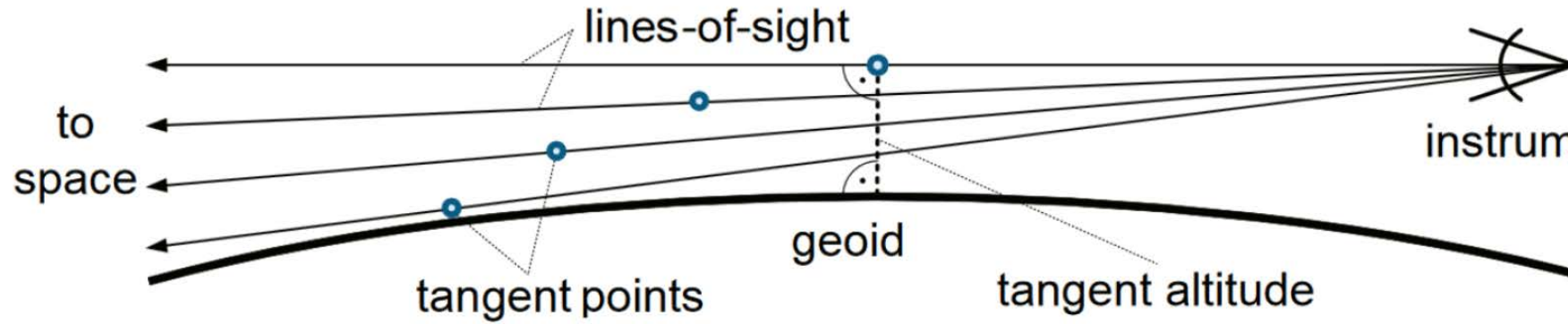
Hamaoui-Laguel et al., 2014

LETTER Kirkby et al., Nature, 2011

doi:10.1038/nature10343

Role of sulphuric acid, ammonia and galactic cosmic rays in atmospheric aerosol nucleation

Infrared limb sounding

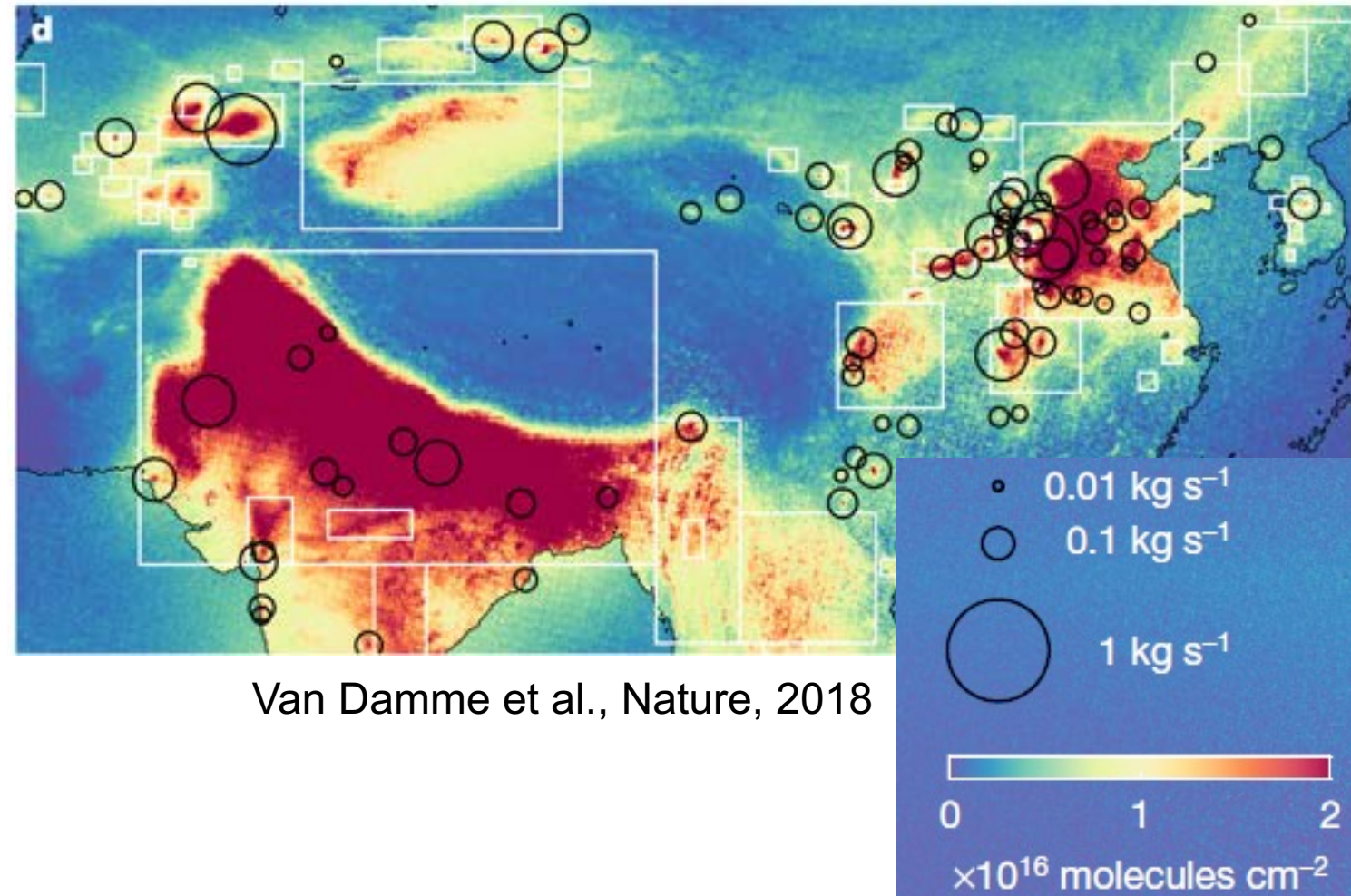


GLORIA@
Geophysica 2017



Observing NH₃

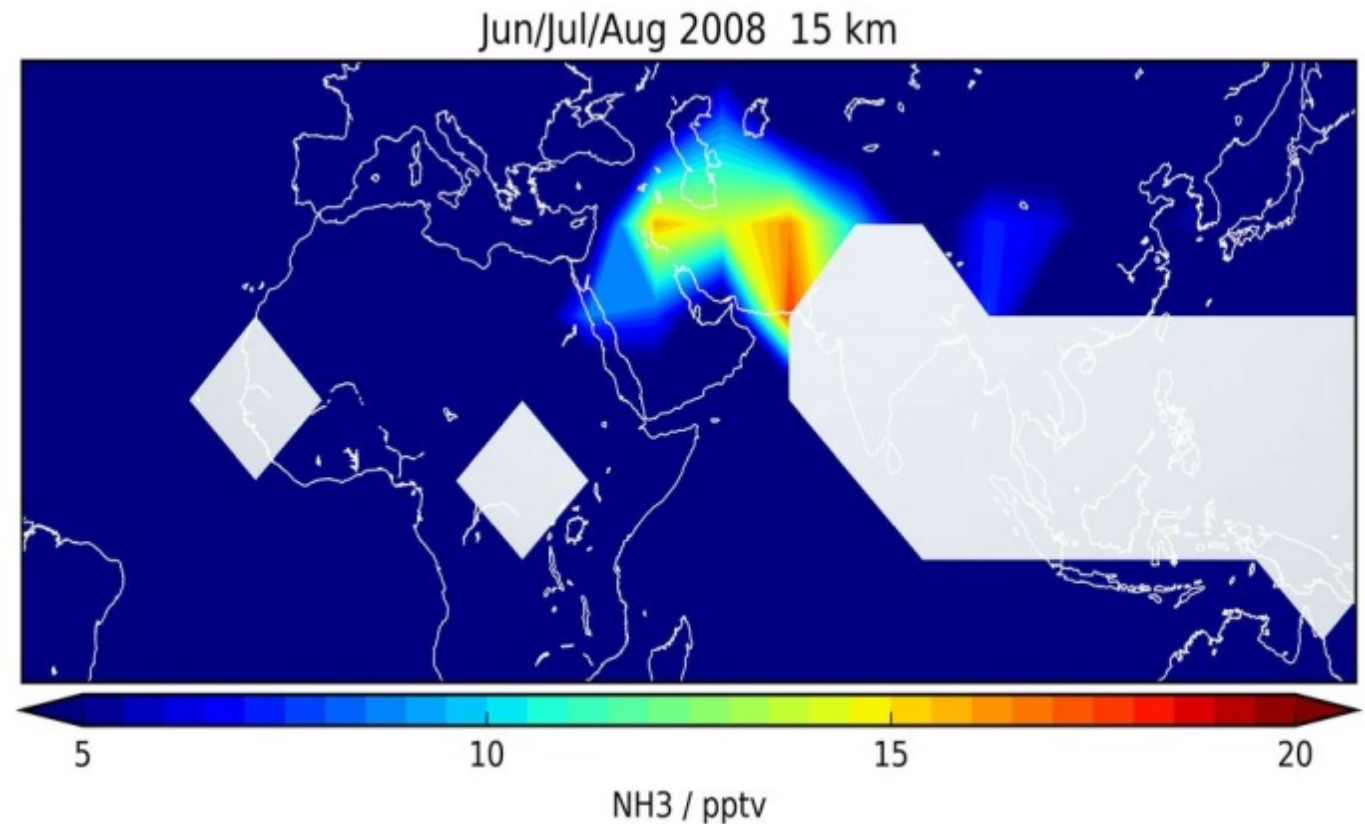
- Ground-based in-situ
- Airborne in-situ up to ~5 km
- Balloon-borne in-situ (no detection above 8 km)
- Ground-based FTIR (columns)
- **Satellite: IR nadir sounding (e.g. IASI)** →
- **For the first time detected in the upper troposphere by IR limb sounding (MIPAS)**



MIPAS-Envisat: NH₃ in the Upper Troposphere

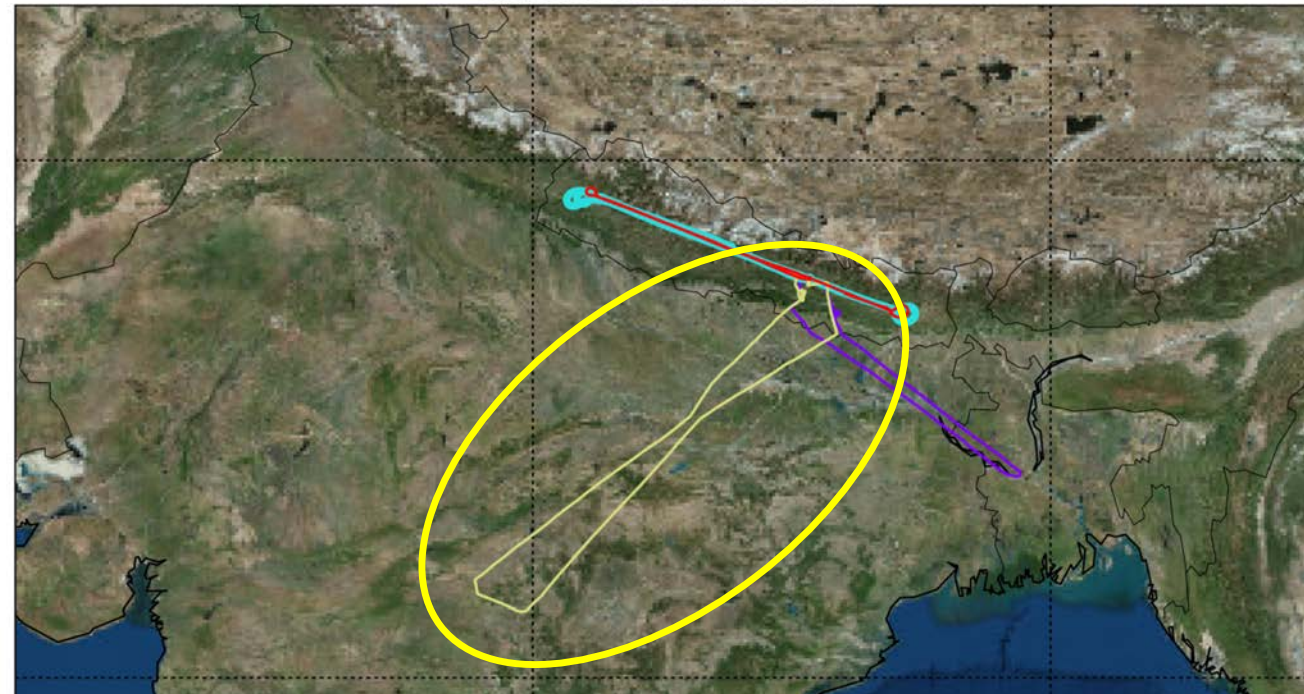


- Evidence for the presence of ammonia in the upper troposphere (Höpfner et al., ACP, 2016)
- Enhanced 3-monthly mean values of up to ~30 pptv within the Asian monsoon upper troposphere



Flights with GLORIA

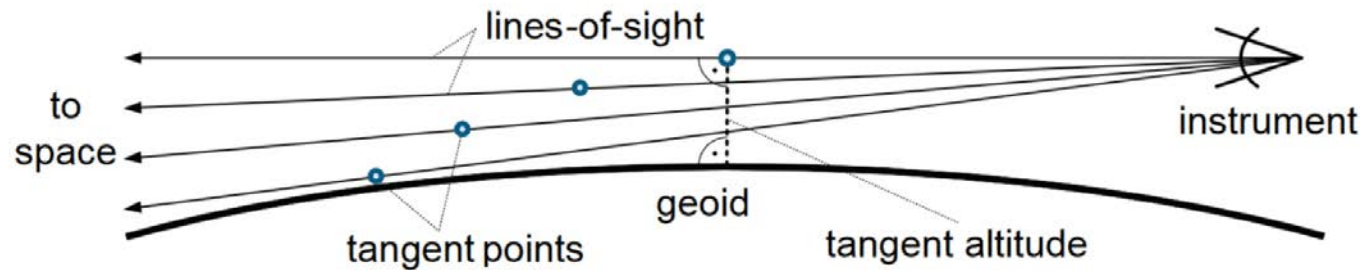
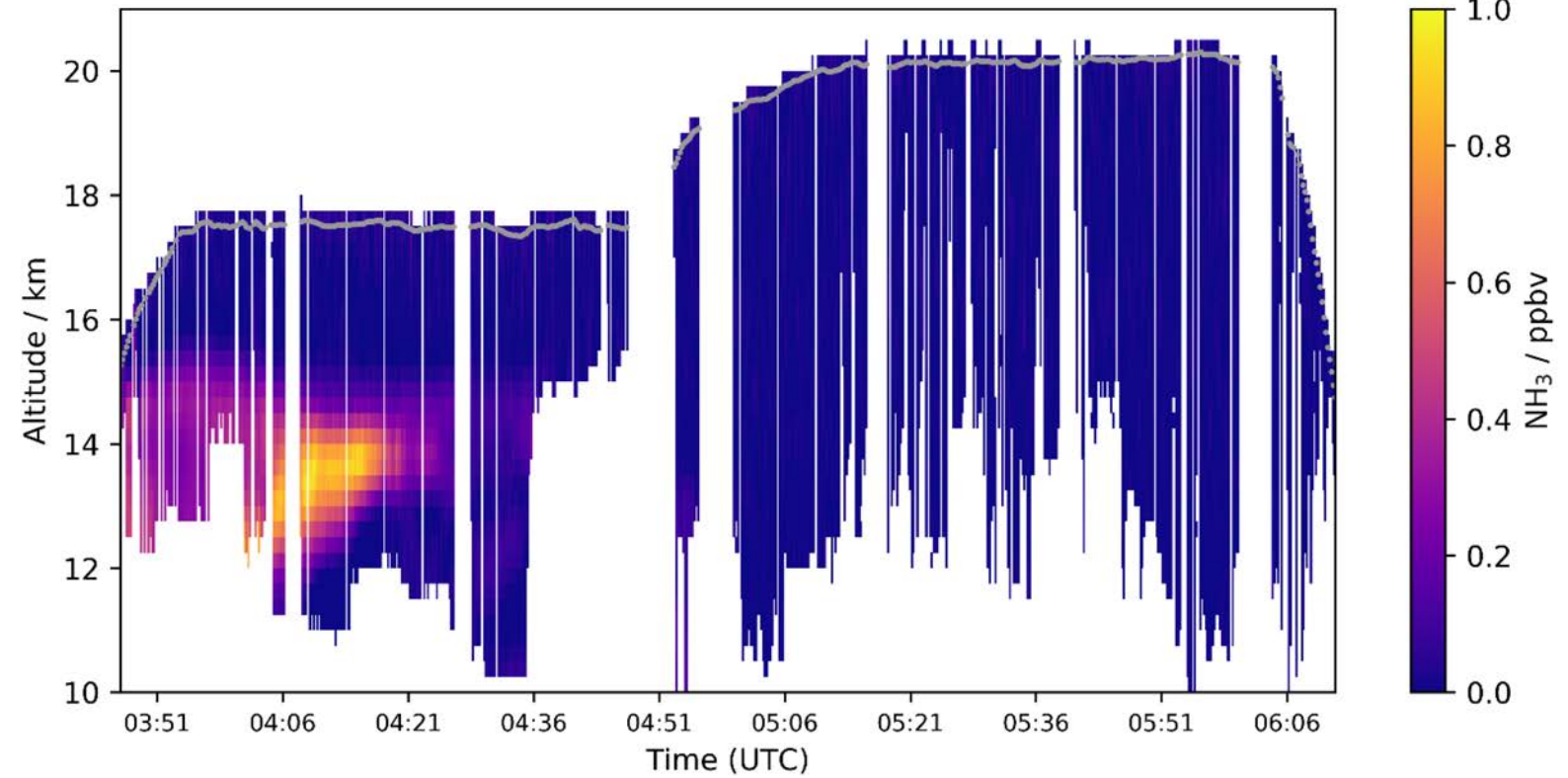
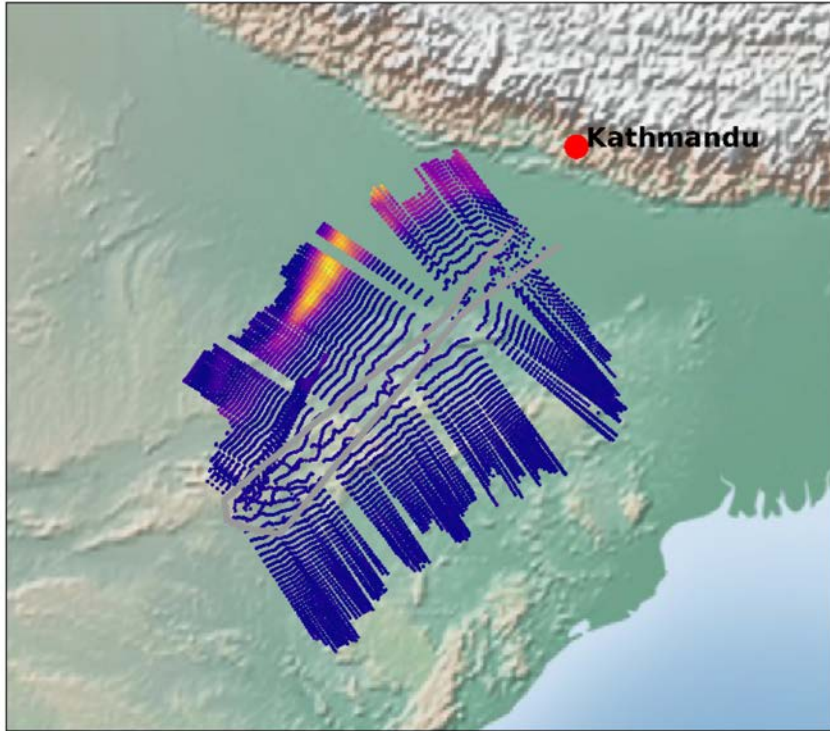
StratoClim GLORIA flight tracks



- Flight 01 2017-07-27
- Flight 02 2017-07-29
- Flight 03 2017-07-31
- Flight 04 2017-08-02

StratoClim flight 31 Jul 2017

NH₃

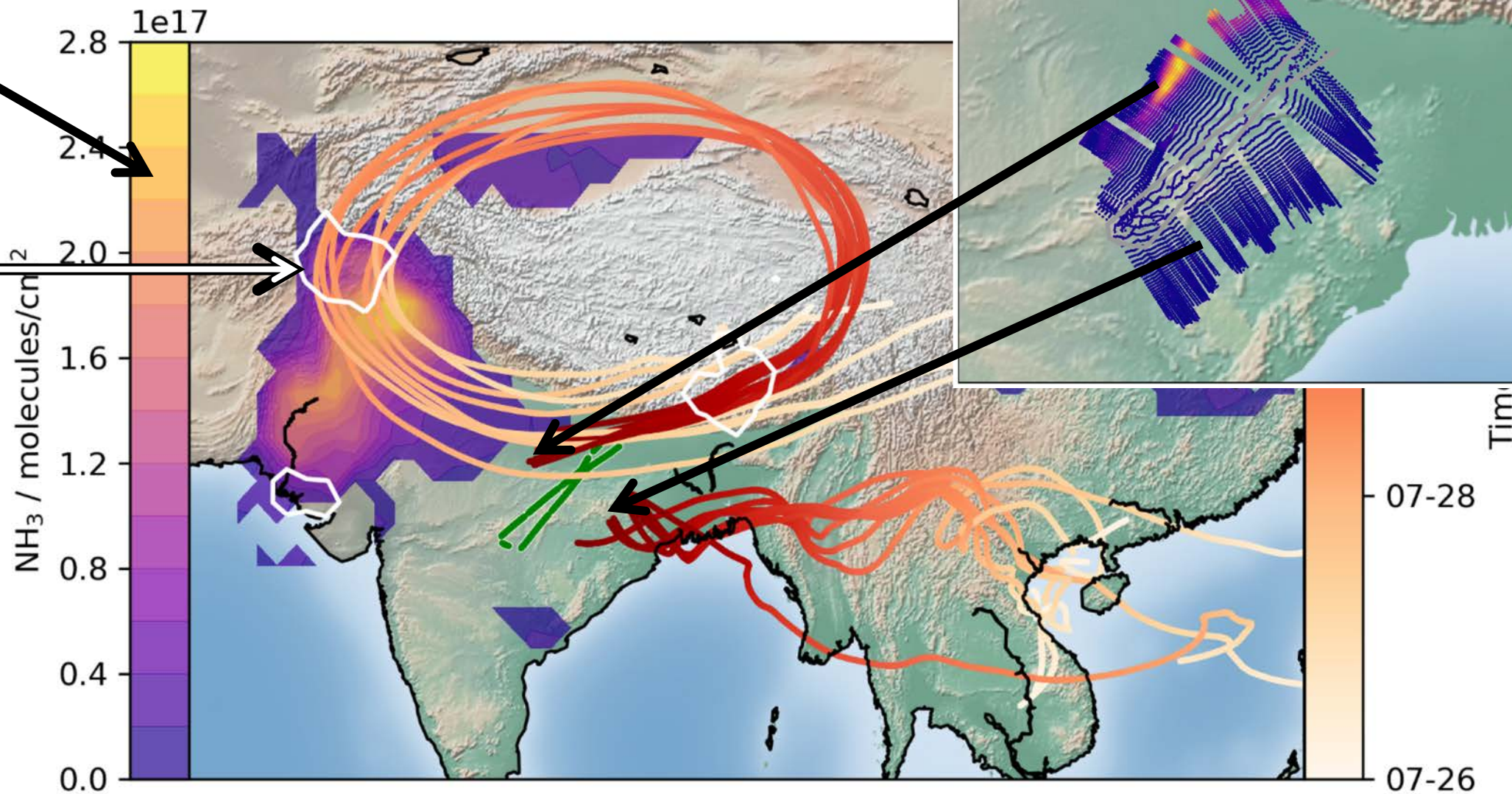


Origin of the NH₃ – plume observed by GLORIA

IASI NH₃
(low-mid
troposphere)

Density of
convective
events along
backward
trajectories

<http://iasi.aeris-data.fr/NH3>
The data produced at
Université Libre de
Bruxelles (ULB) by Simon
Whitburn and Martin Van
Damme
Van Damme et al., AMT,
2017

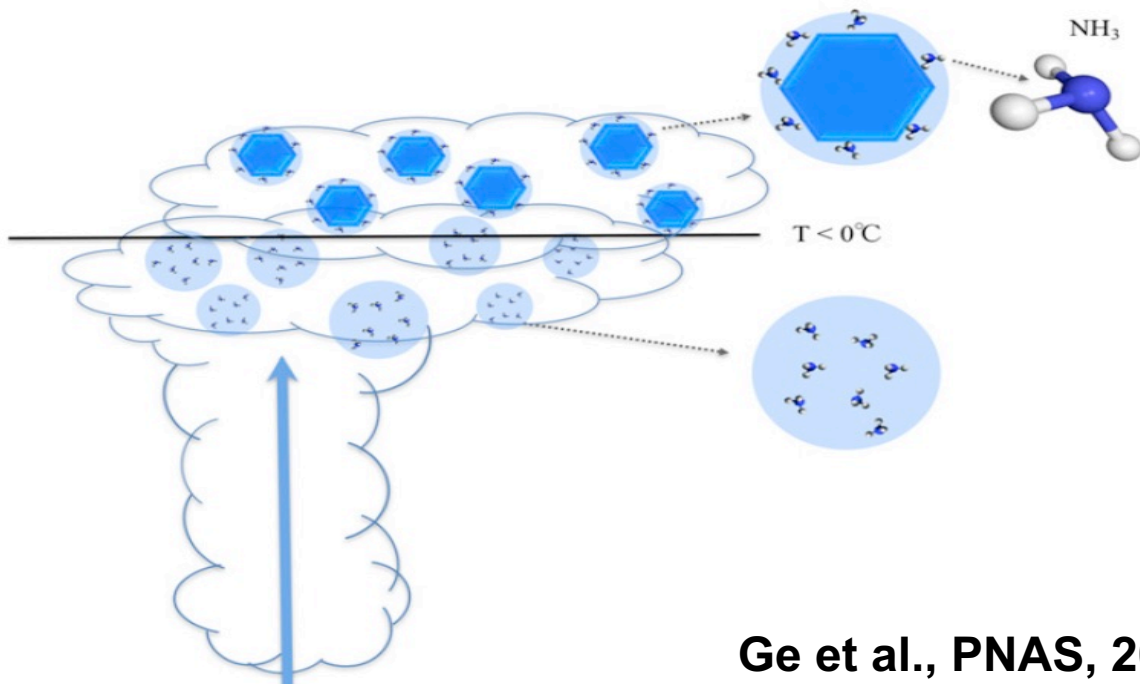


Why is NH_3 not washed out during convection?

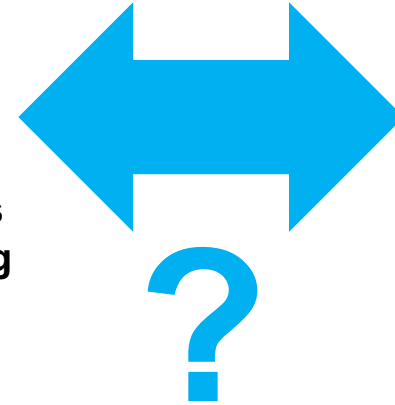
A molecular perspective for global modeling of upper atmospheric NH_3 from freezing clouds

Cui Ge^{a,1}, Chongqin Zhu^{b,1}, Joseph S. Francisco^{b,2}, Xiao Cheng Zeng^{b,2}, and Jun Wang^{a,2}

- Study trying to explain the MIPAS NH_3 observations
- “We show that the NH_3 dissolved in liquid cloud droplets is prone to being released into the UTLS upon freezing during deep convection.”



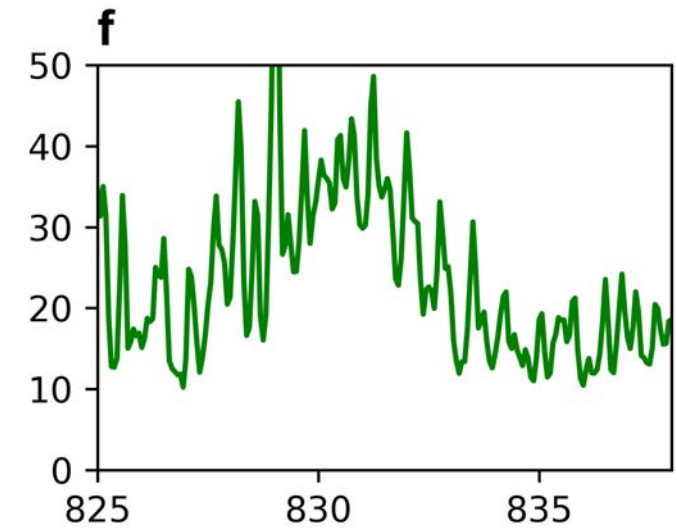
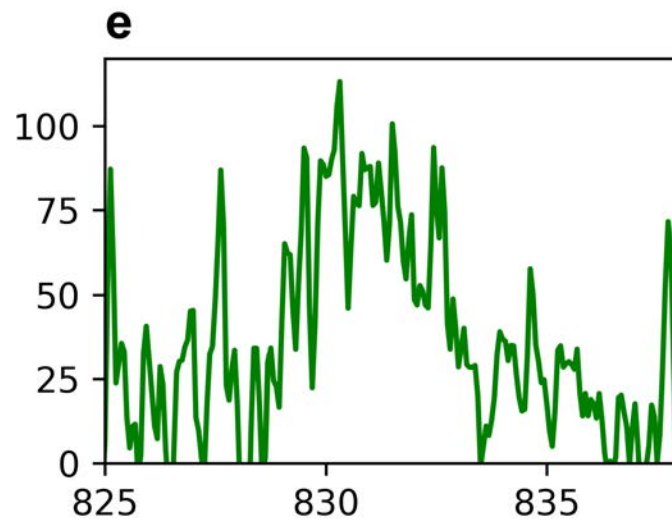
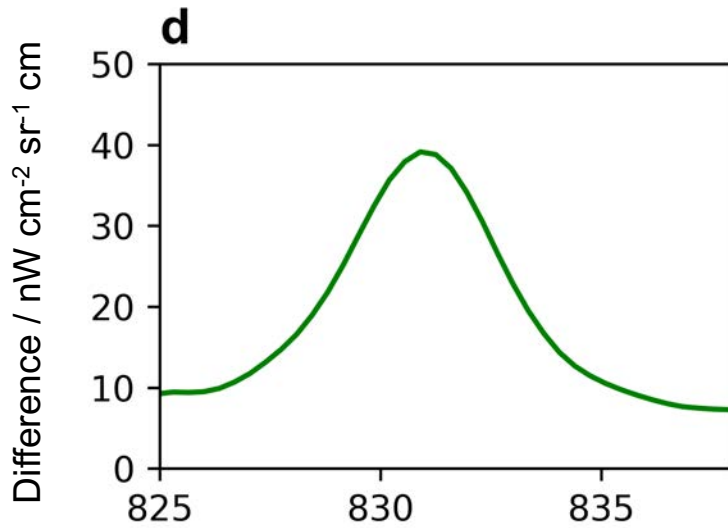
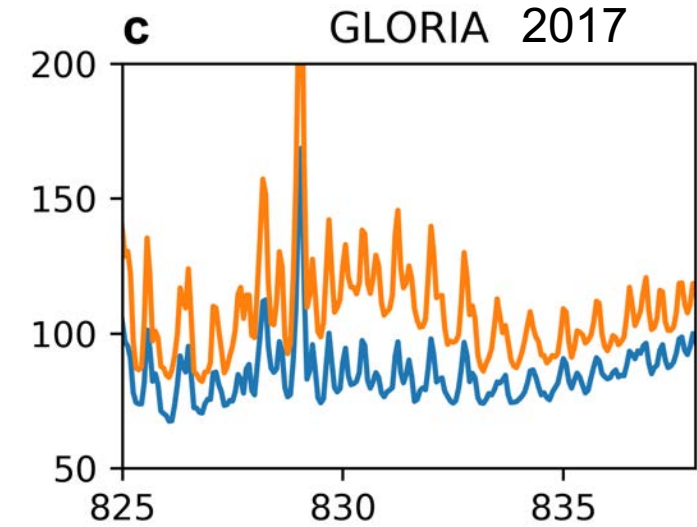
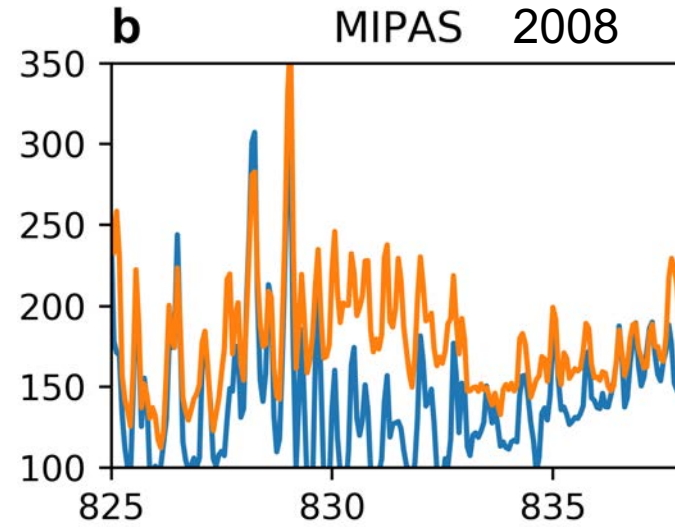
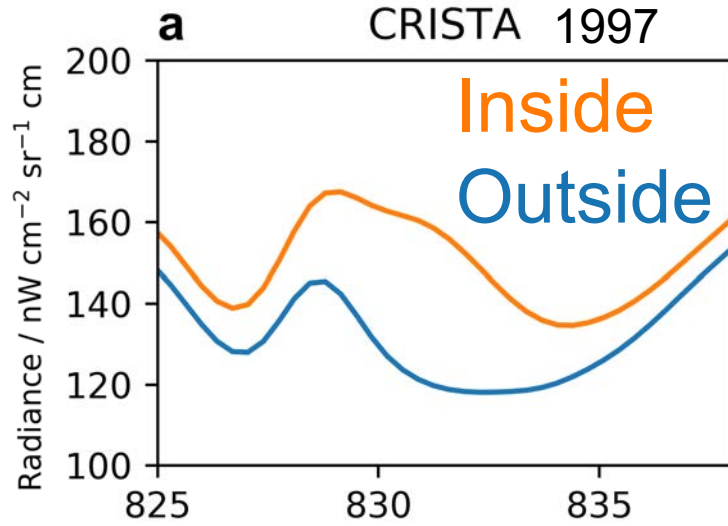
Ge et al., PNAS, 2018



- ph – dependence of NH_3 solubility in liquid water:
„Convective clouds are hardly acidic so that NH_3 is only partly dissolved and removed by precipitation“

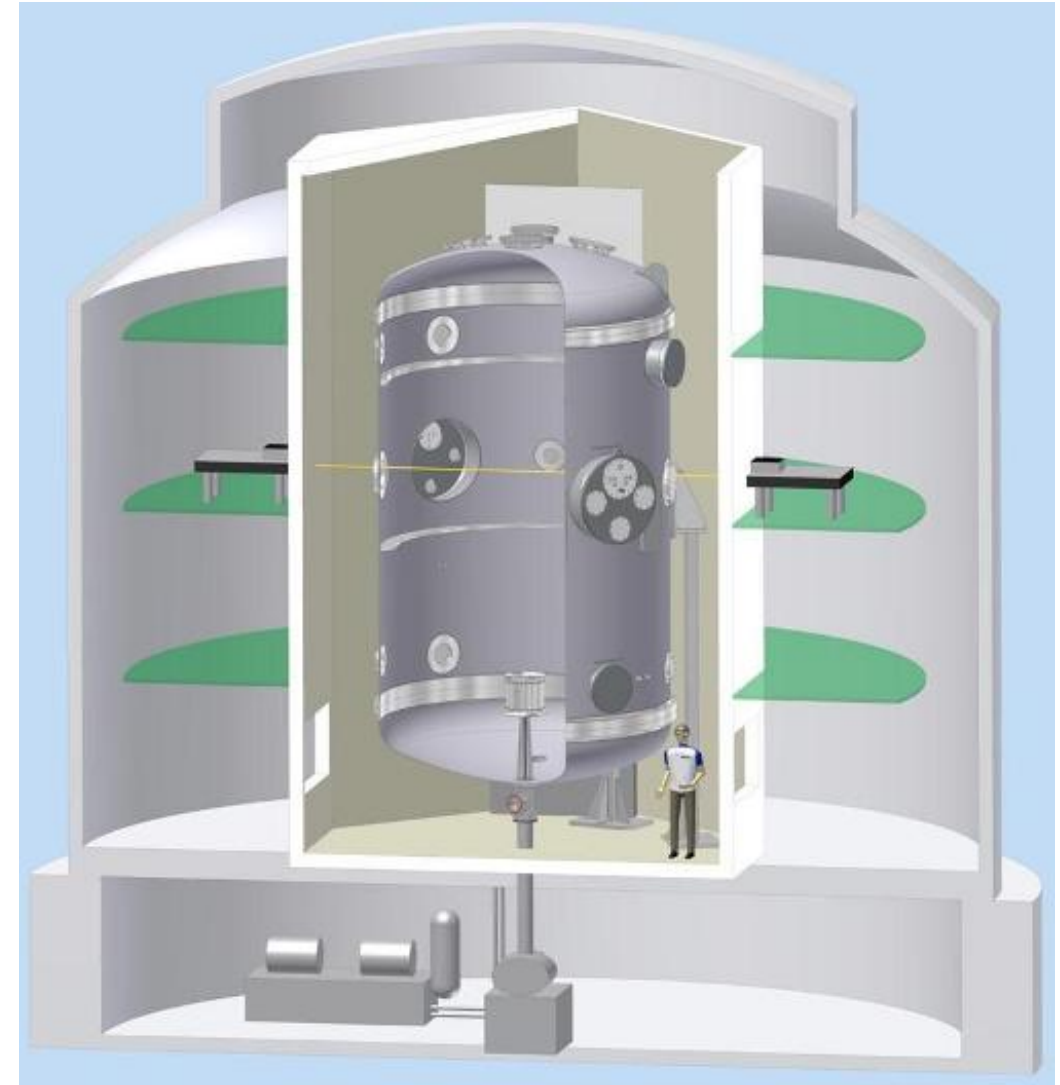
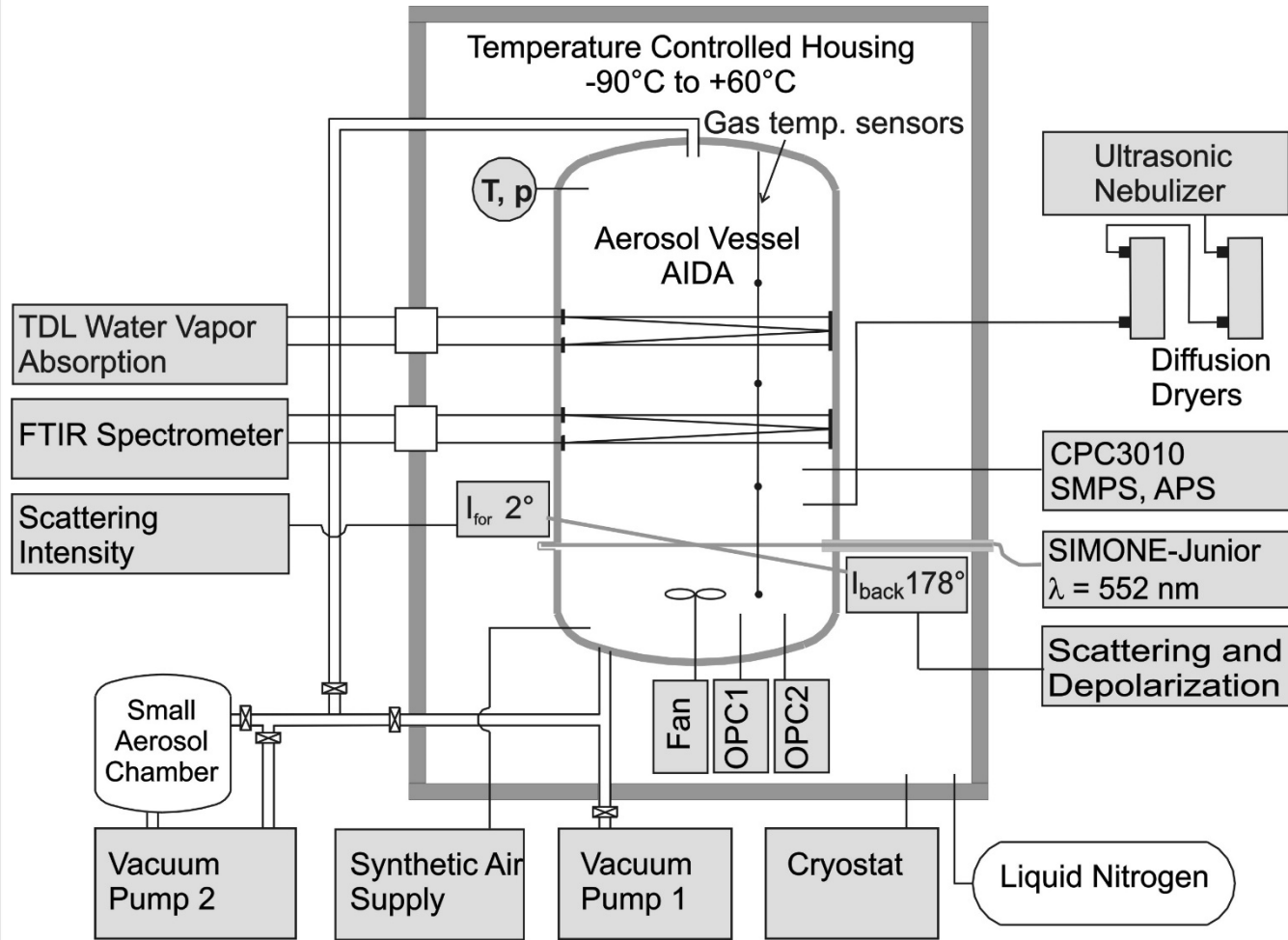
Metzger et al., JGR, 2002

A peak in infrared spectra inside the monsoon upper troposphere at 831 cm^{-1}

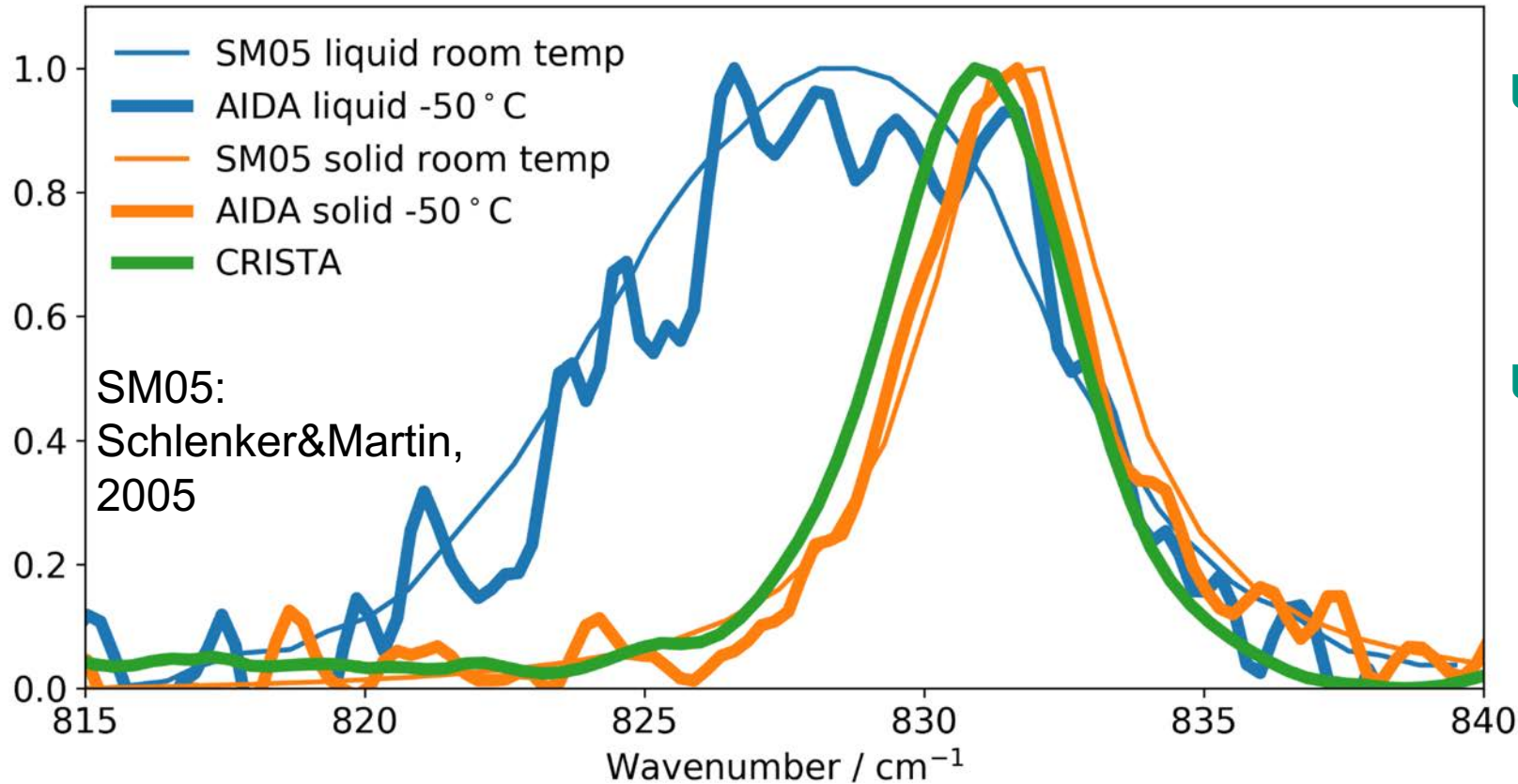


Wavenumber / cm^{-1}

The AIDA aerosol and cloud chamber



Laboratory infrared spectra of the $\nu_2(\text{NO}_3^-)$ band of NH_4NO_3 particles compared to the observations

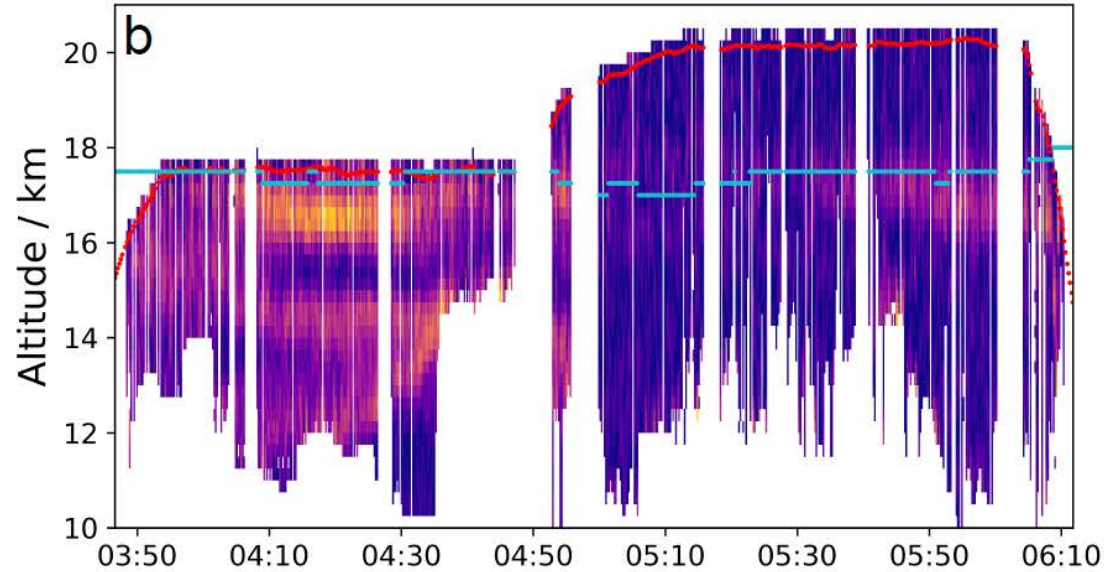
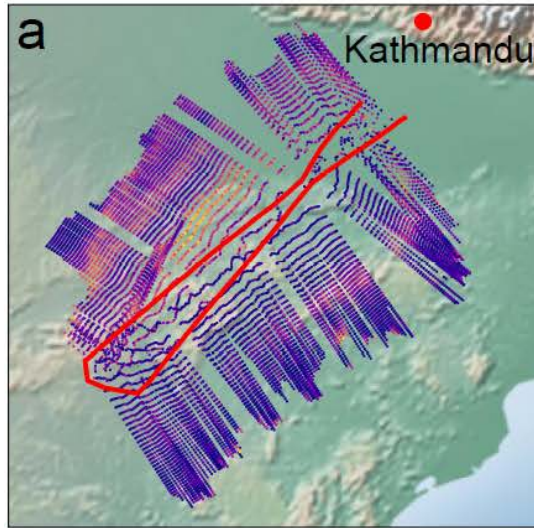


- Observed peak fits to the signature of **solid ammonium nitrate**
- **Liquid ammonium nitrate** has a much broader peak and **does not fit the observations**

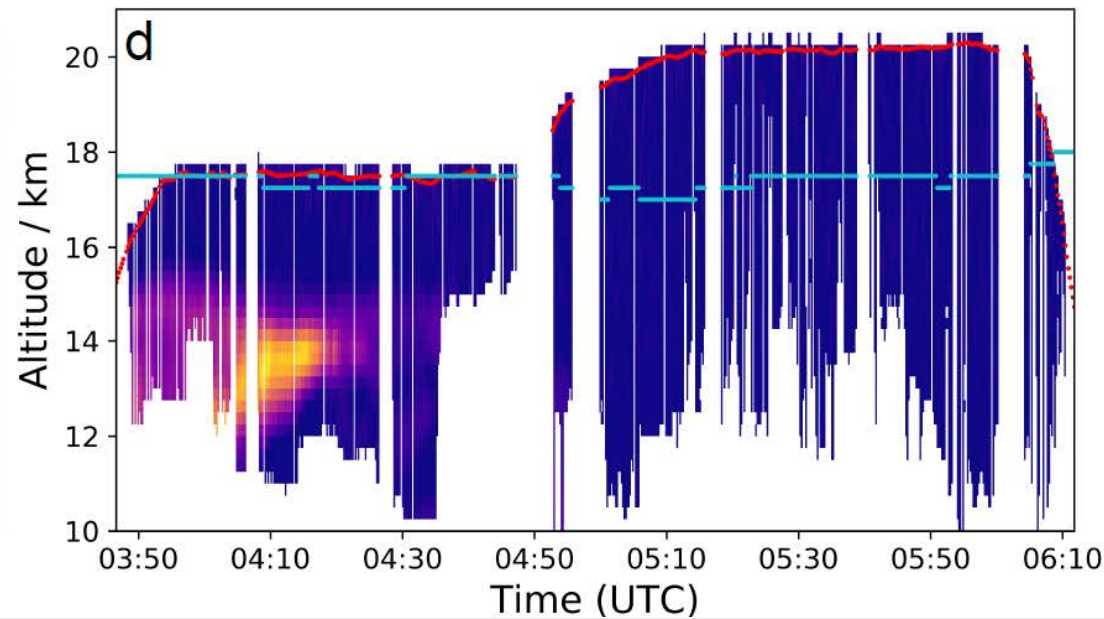
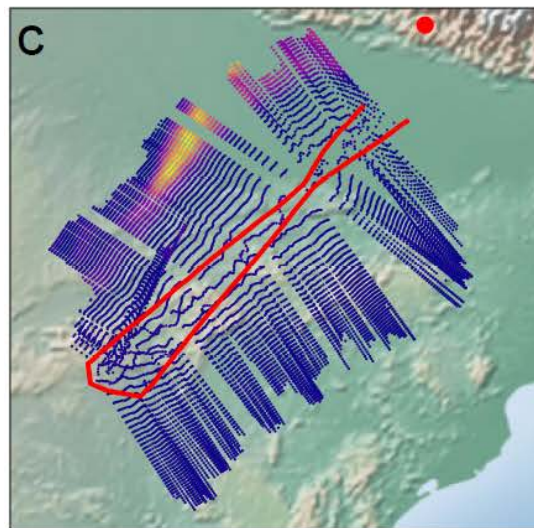


AIDA observations allow to derive ammonium nitrate mass concentration profiles from the infrared limb observations

StratoClim flight 31 Jul 2017

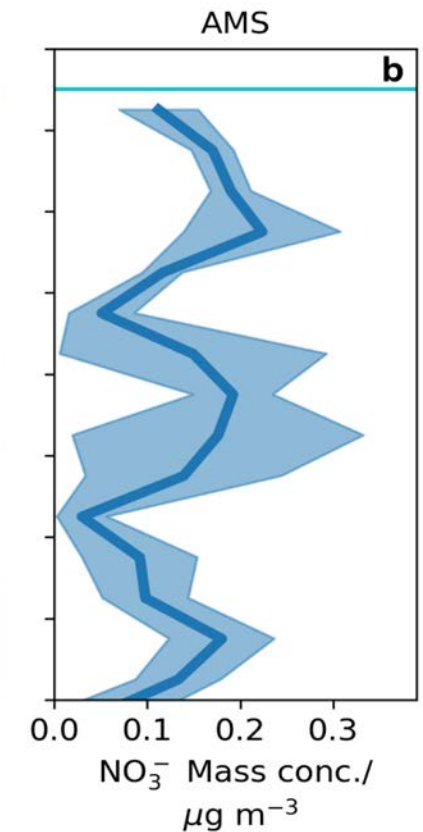
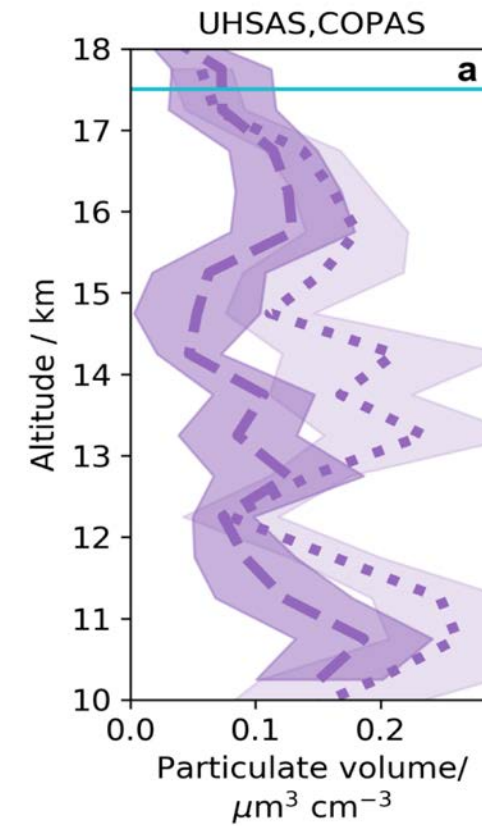
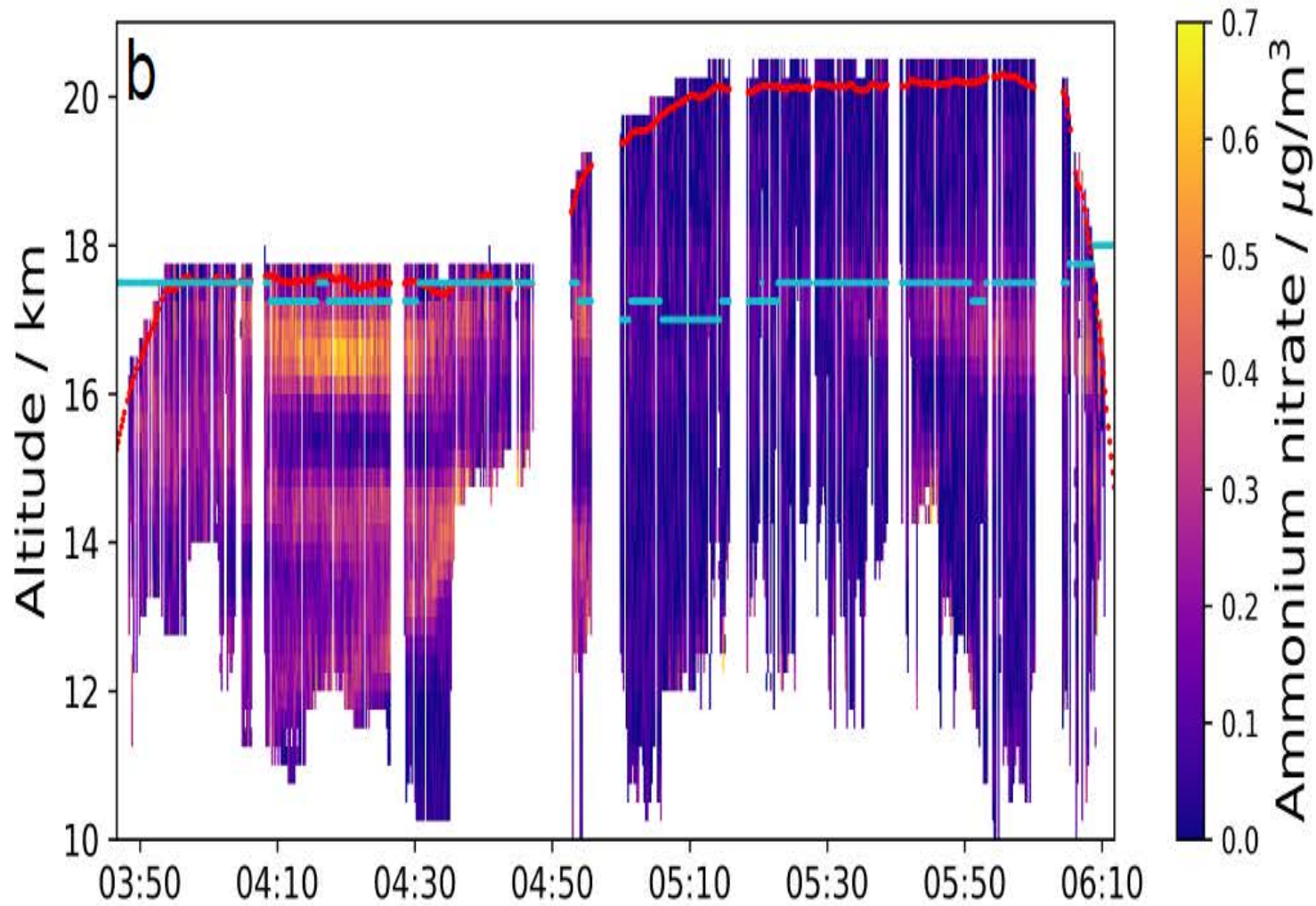


**NH_4NO_3
aerosol
mass density**

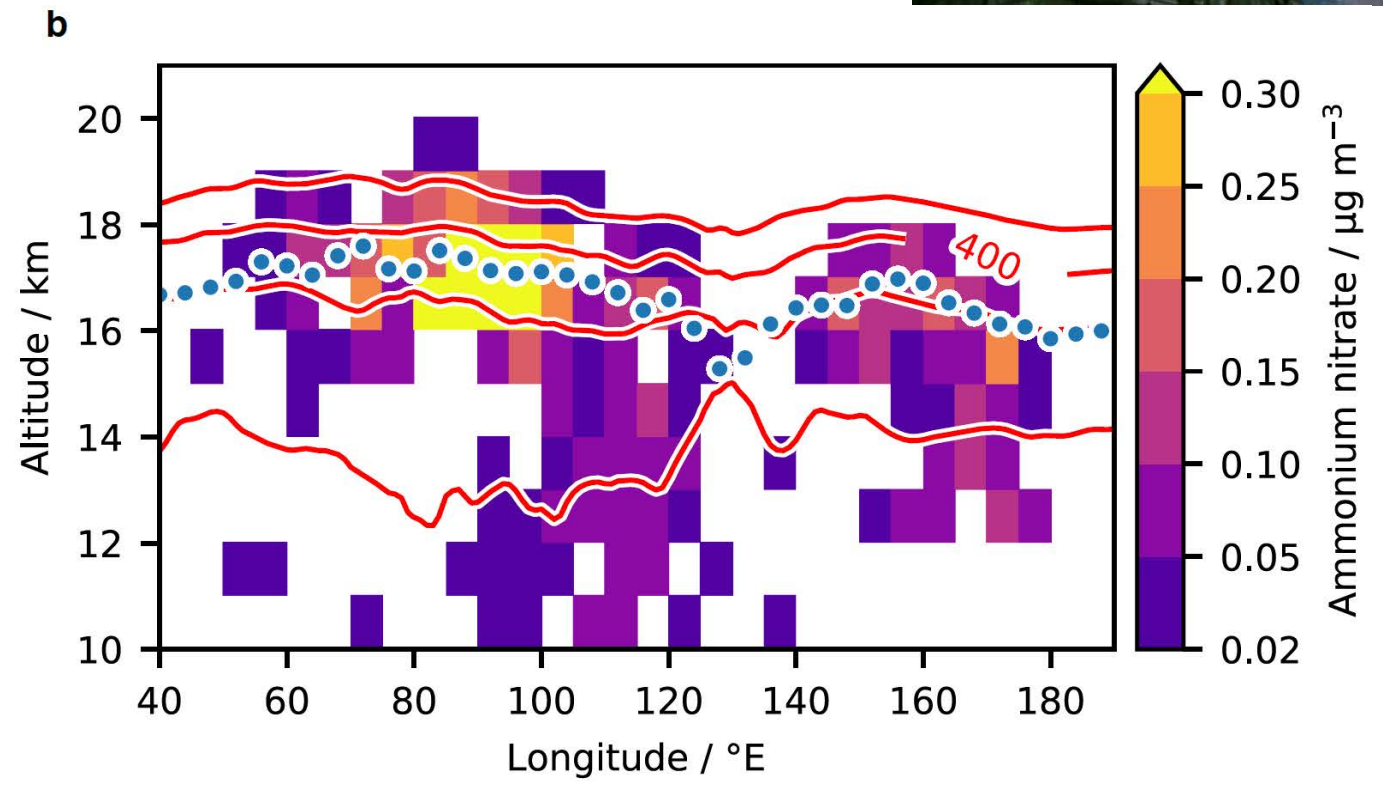
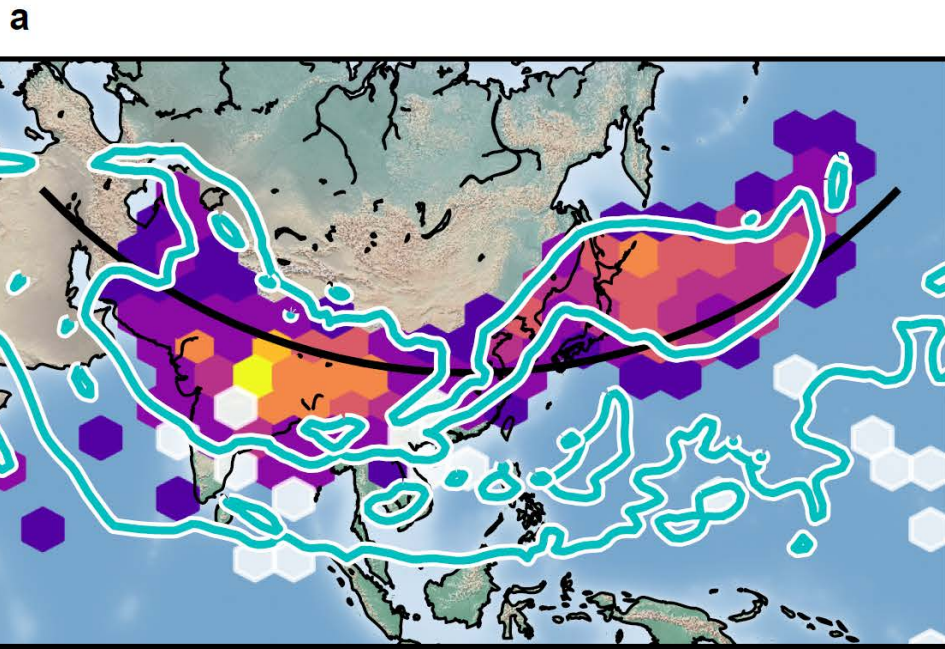
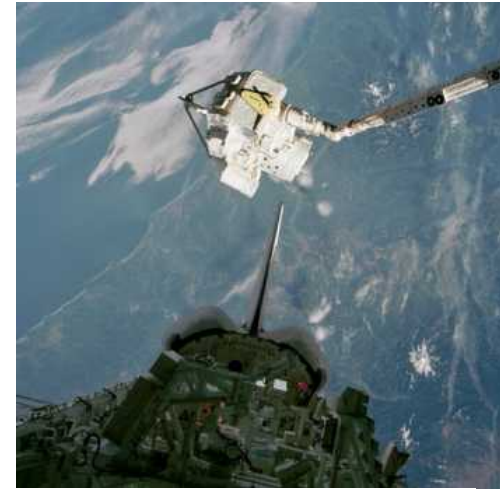


**NH_3
trace gas
mixing ratio**

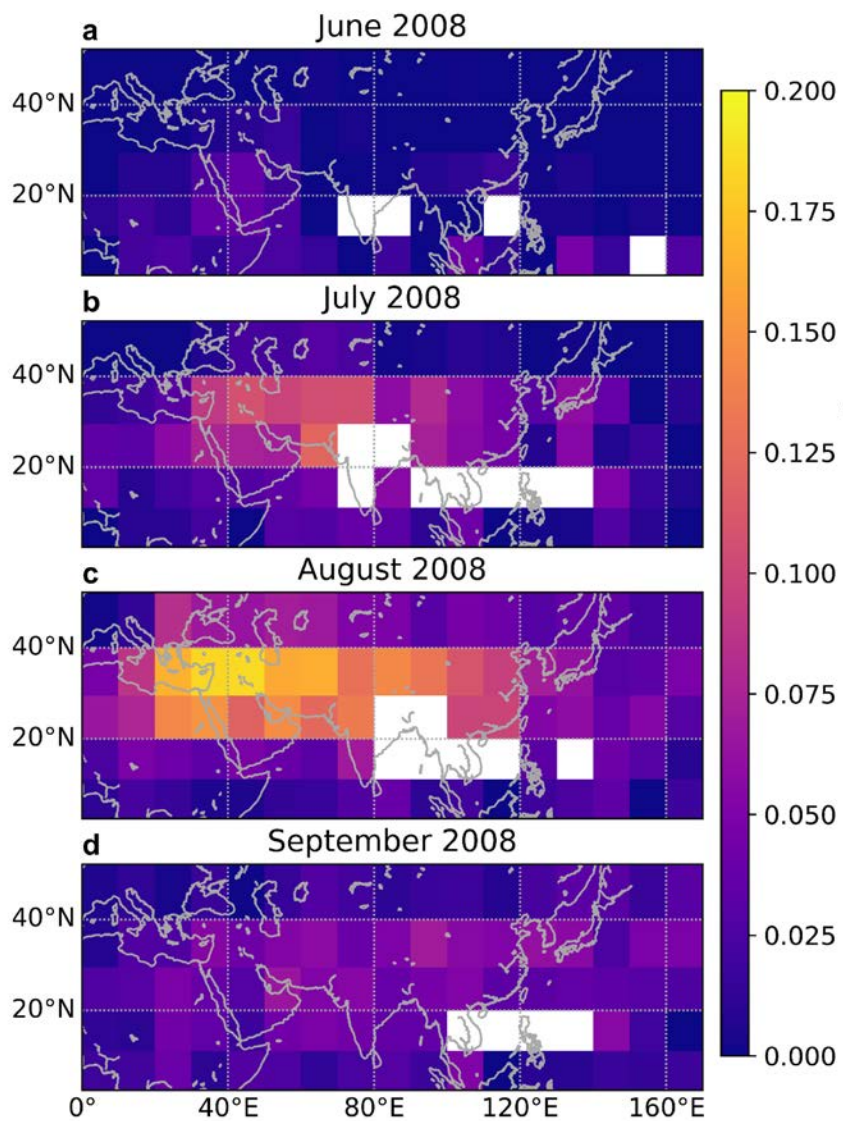
StratoClim flight 31 Jul 2017: comparison with in-situ aerosol measurements of Univ./MPI Mainz



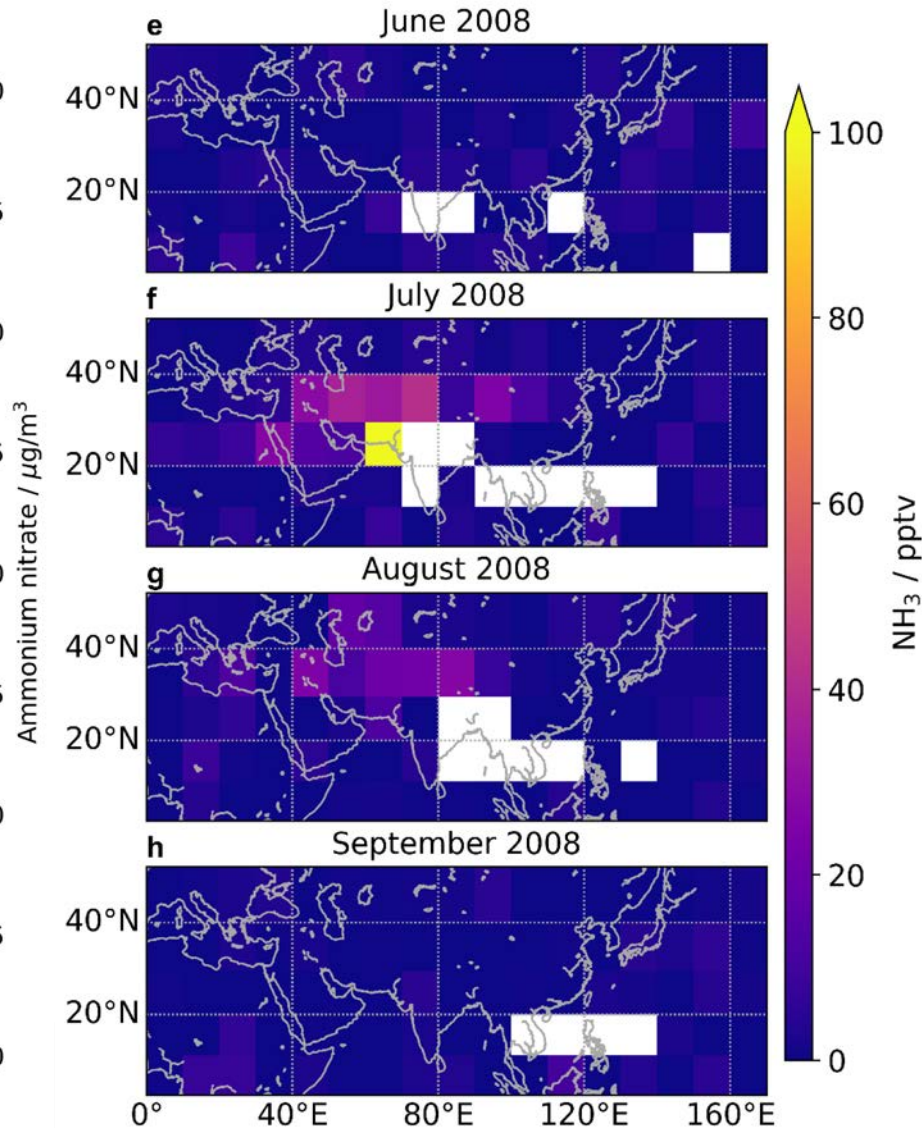
CRISTA ammonium nitrate August 1997



NH_4NO_3 aerosol mass density



NH_3 trace gas mixing ratio



MIPAS 2008

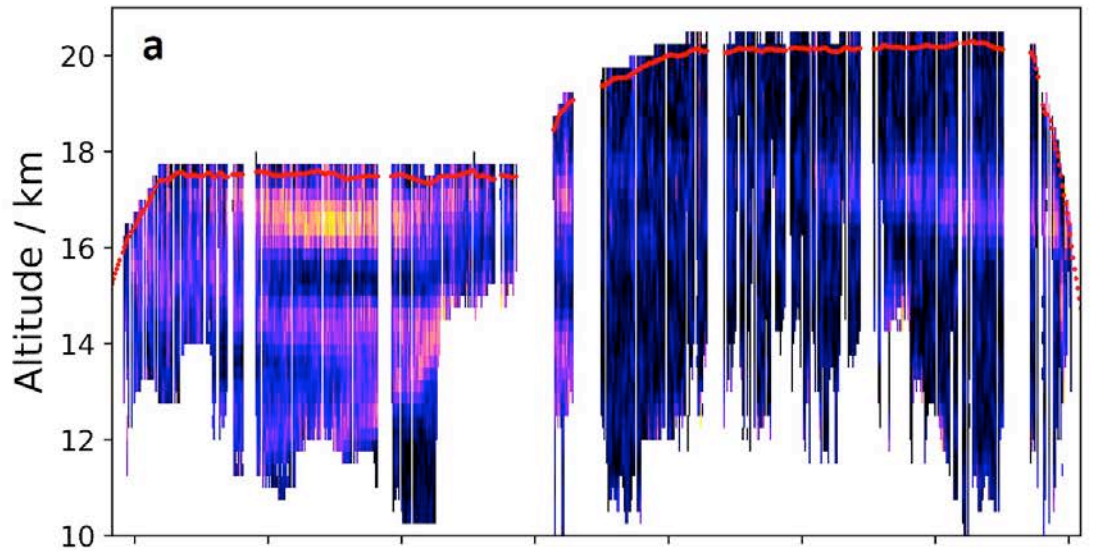
Formation and possible impact of solid ammonium nitrate in the UT

- Cziczo & Abbatt, 2000: **NH_4NO_3 shows strong inhibition to efflorescence down to 2% RH (298 – 238 K)** “*These findings strongly suggest that, in the absence of heterogeneous nuclei, a wide variety of inorganic aerosols will exist as liquid solutions in the atmosphere regardless of relative humidity and temperature conditions*”
- Abbatt et al., Science, 2006: “**Solid Ammonium Sulfate Aerosols as Ice Nuclei: A Pathway for Cirrus Cloud Formation**” (Laboratory and model study)
- Our AIDA experiments: **solid NH_4NO_3 (223 K) only forms in presence of small impurities (3 mol%) of ammonium sulfate**

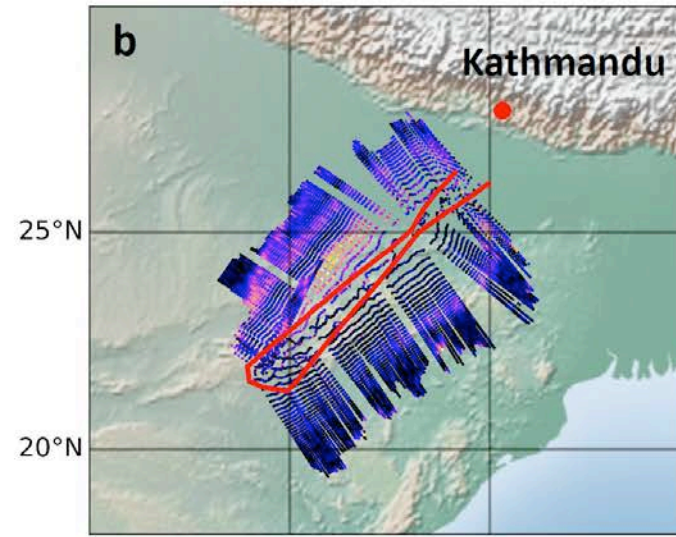
Summary

- Observations of **NH₃ concentrations > 1 ppbv in the upper troposphere** during StratoClim in Jul/Aug 2017
- NH₃ source region: Pakistan/NW India, **upward transport by convection**
- Detection of spectral signal of **solid ammonium nitrate** aerosol particles in limb infrared spectra of CRISTA, MIPAS and GLORIA and in IR absorption spectra in AIDA
- **NH₄NO₃ profiles** retrieved from limb-observations by use of **IR mass absorption coefficients as determined in AIDA**
- **NH₄NO₃ aerosols prevalent in the Asian monsoon anticyclone** following enhanced values of NH₃: evidence that the Asian tropopause aerosol layer (ATAL) consists (partly) of ammonium nitrate
- **Solid NH₄NO₃ particle formation** in AIDA appears only in the presence of **impurities of (NH₄)₂SO₄** - these may act as a **good ice nuclei**

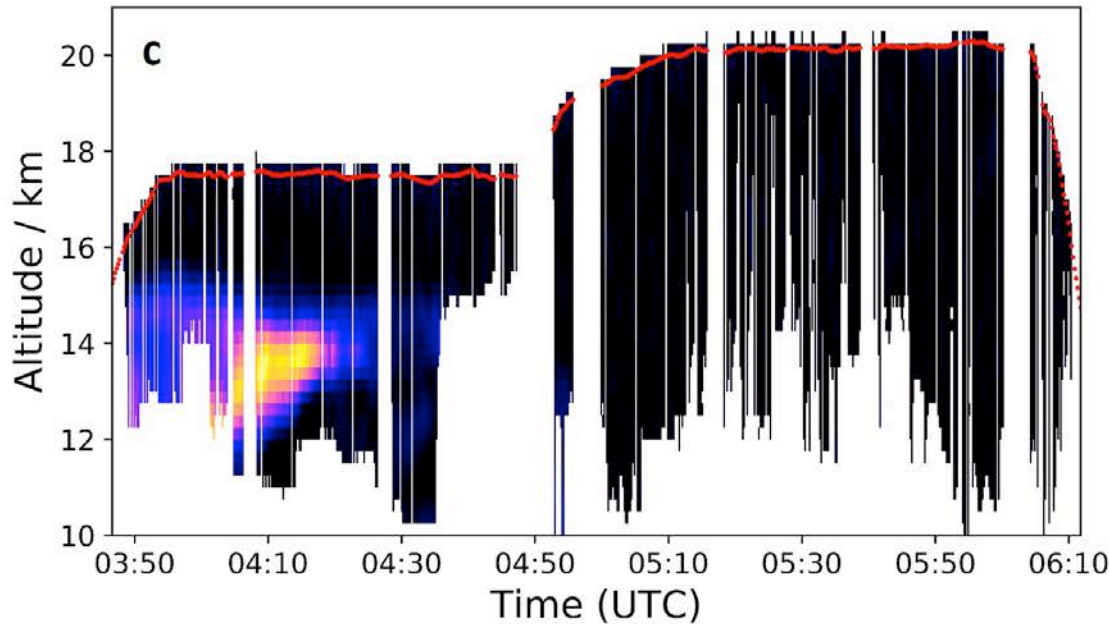
StratoClim flight 31 Jul 2017



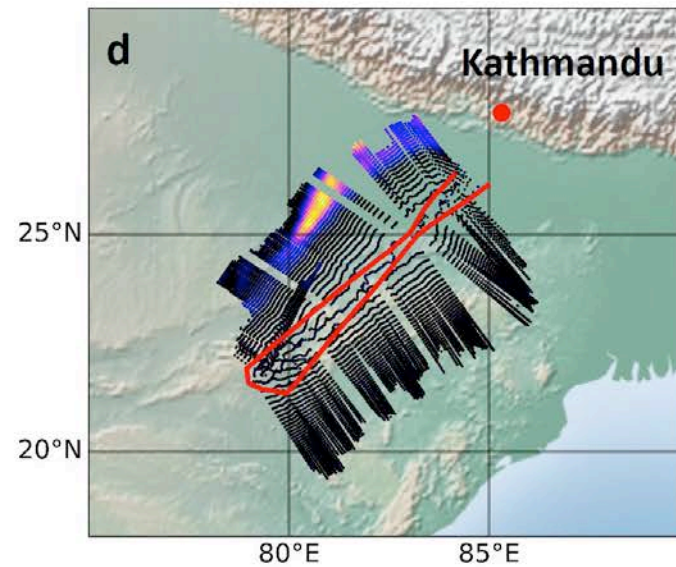
Ammonium nitrate / $\mu\text{g}/\text{m}^3$



**NH_4NO_3
aerosol
mass density**

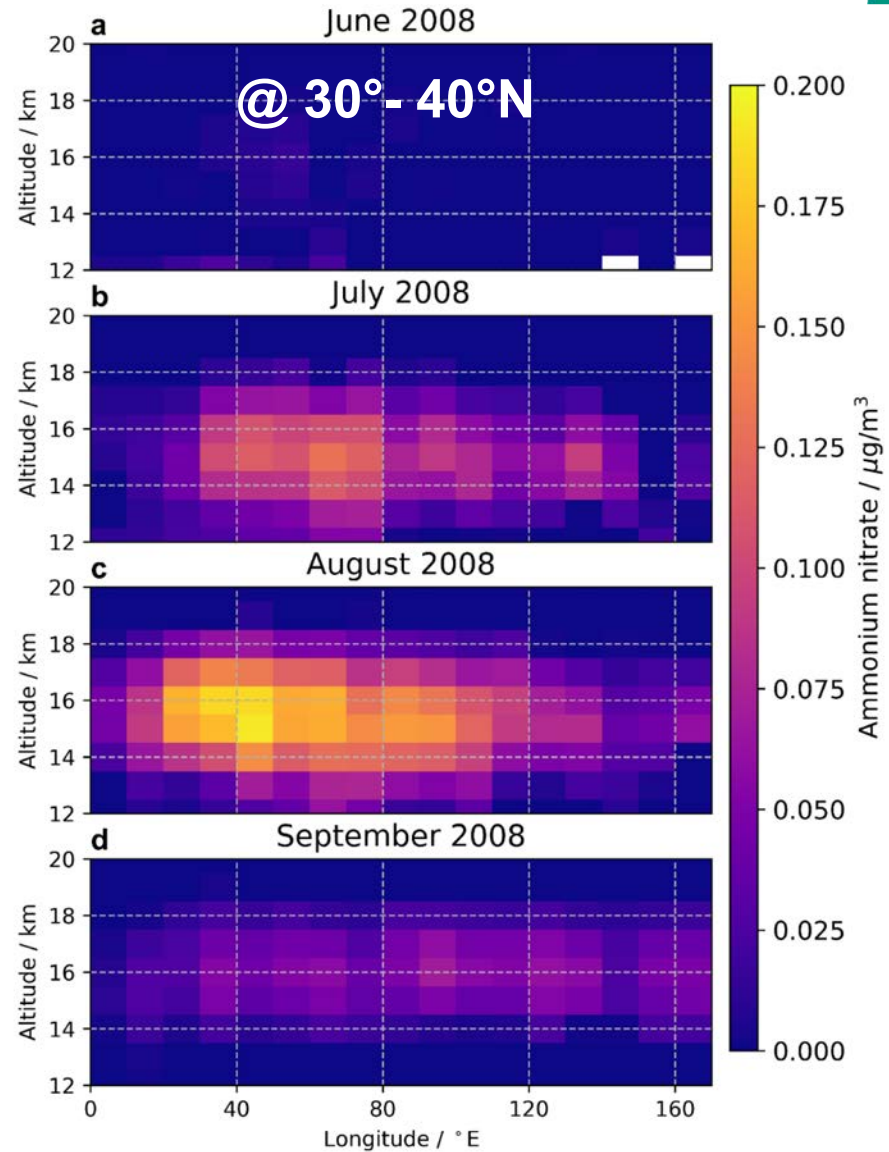
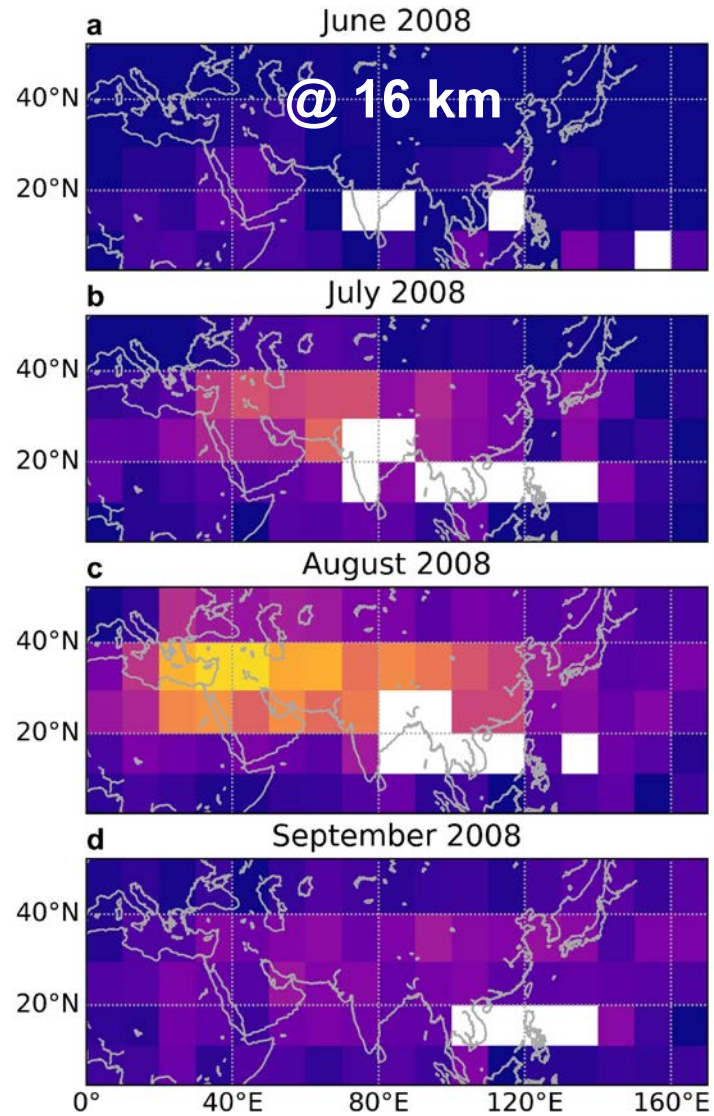


NH_3 / ppbv



**NH_3
trace gas
mixing ratio**

MIPAS 2008: ammonium nitrate



Infrared spectroscopy of ammonium nitrate

The $\nu_2(\text{NO}_3^-)$ band of NH_4NO_3 has been assigned in laboratory spectra to wavenumbers around 831 cm^{-1} :

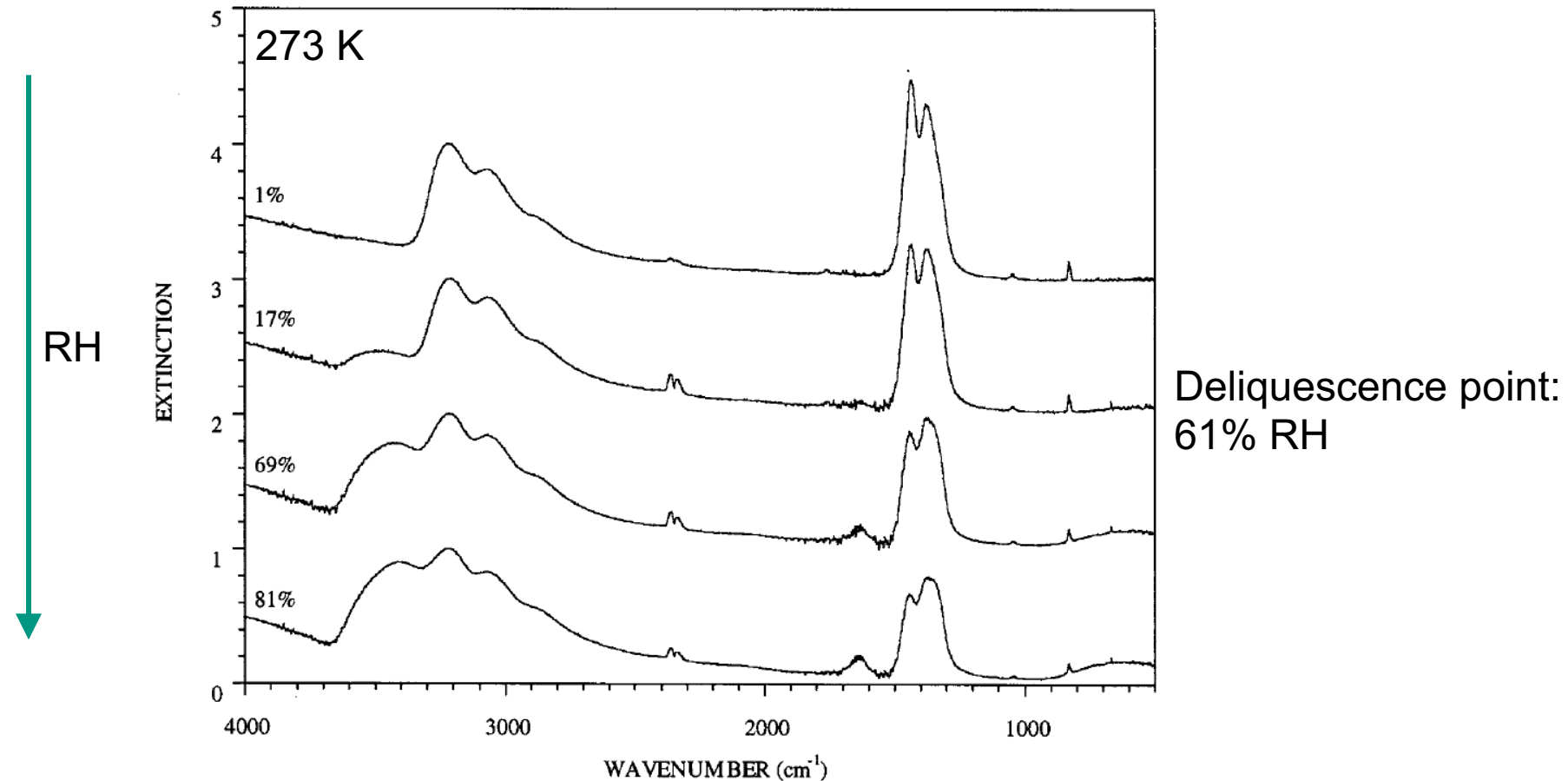
- Théorêt and Sandorfy (1964): 830 cm^{-1} (phase IV)
- Fernandes et al. (1979): 833 cm^{-1} (phase V), 831 cm^{-1} (phase IV)
- Allen et al., 1994: $825\text{-}835 \text{ cm}^{-1}$
- Koch et al. (1996): 832 cm^{-1} (phase V), 830 cm^{-1} (phase IV)
- Schlenker and Martin (2005): 831 cm^{-1}

Literature overview

Cziczo & Abbatt (2000)

NH_4NO_3 : strong inhibition to efflorescence down to 2% RH (298 – 238 K)

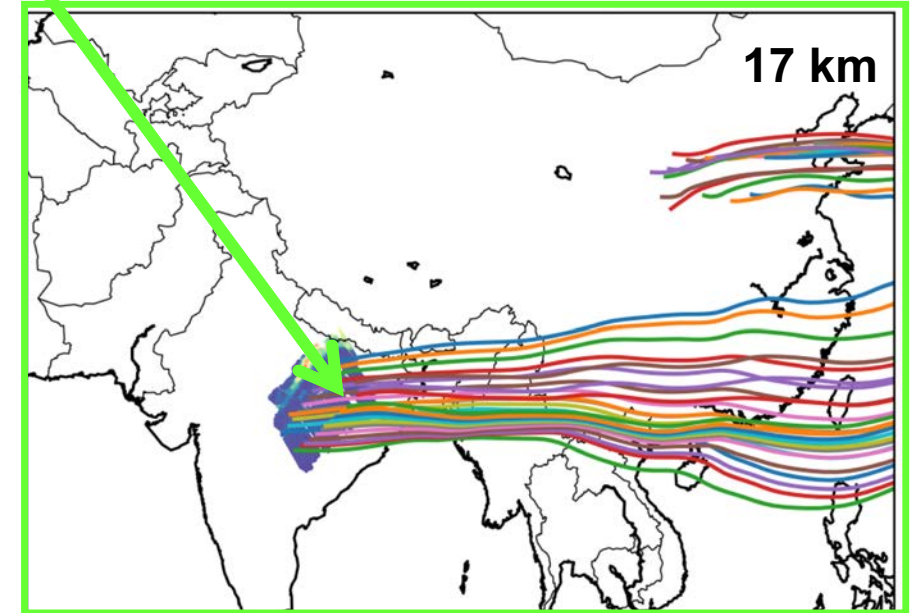
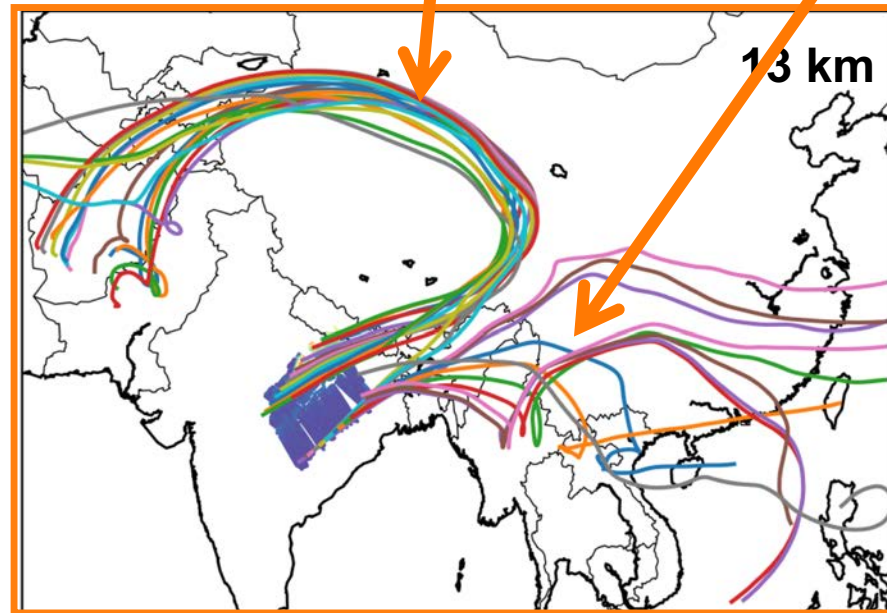
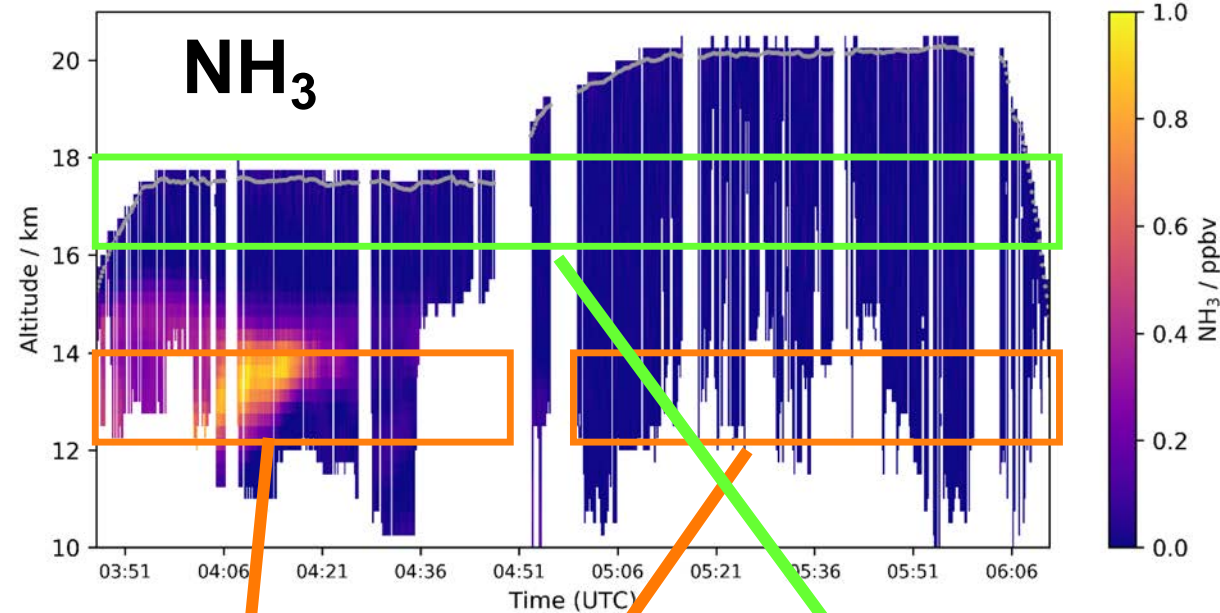
Aerosol flow tube equipped with FTIR spectrometer



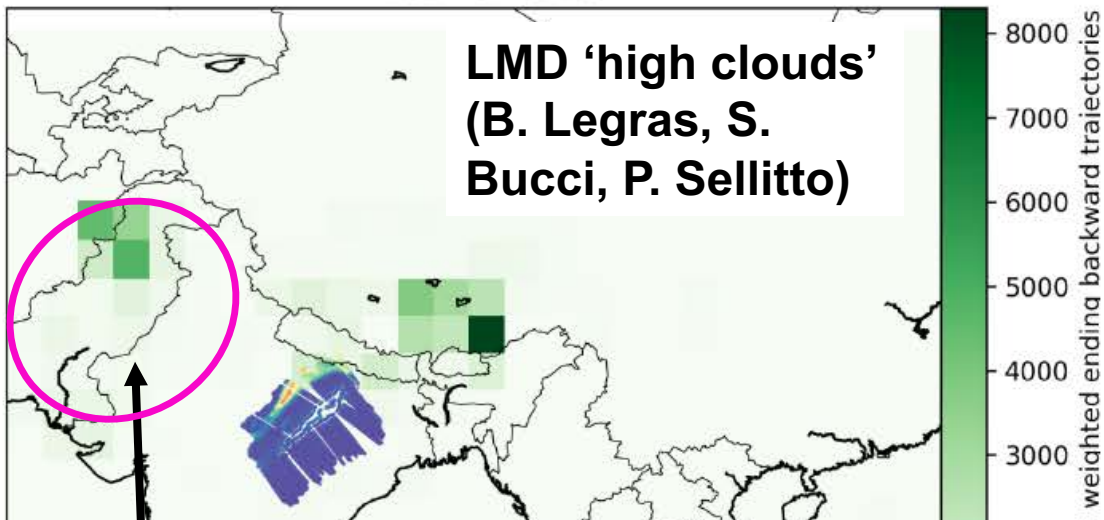
Backtrajectories

2017-07-31

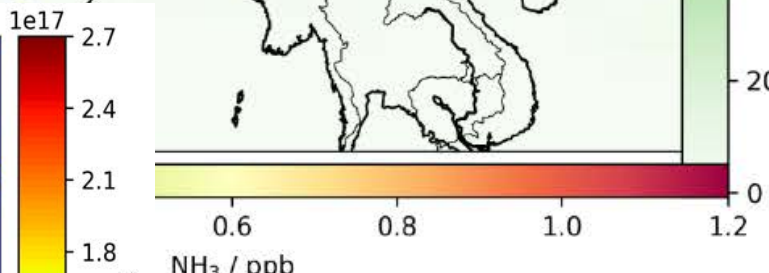
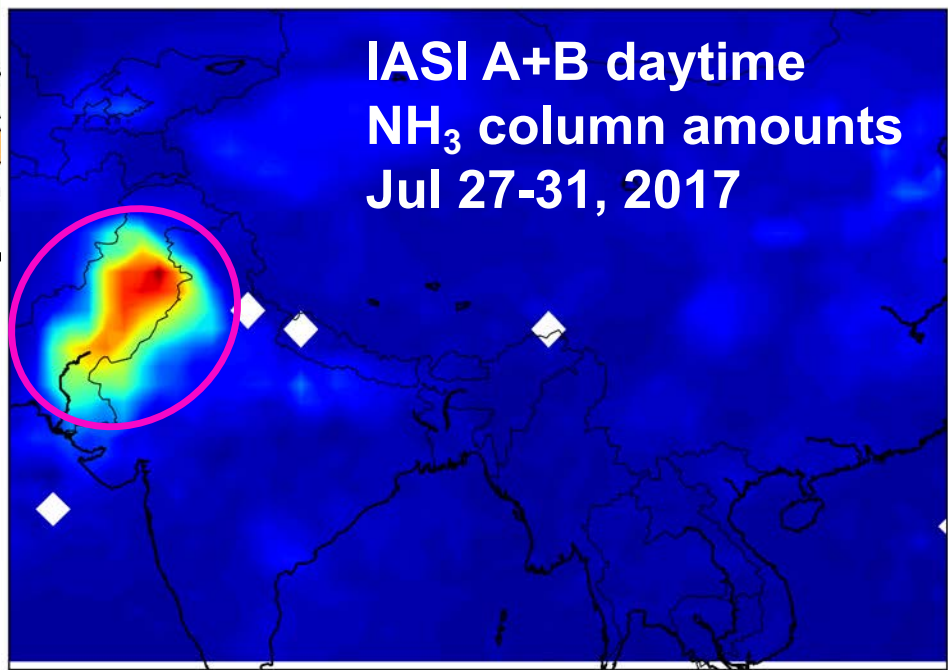
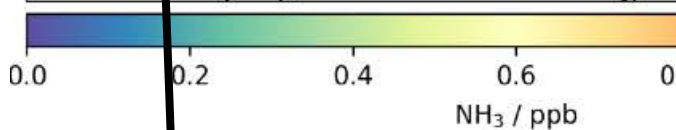
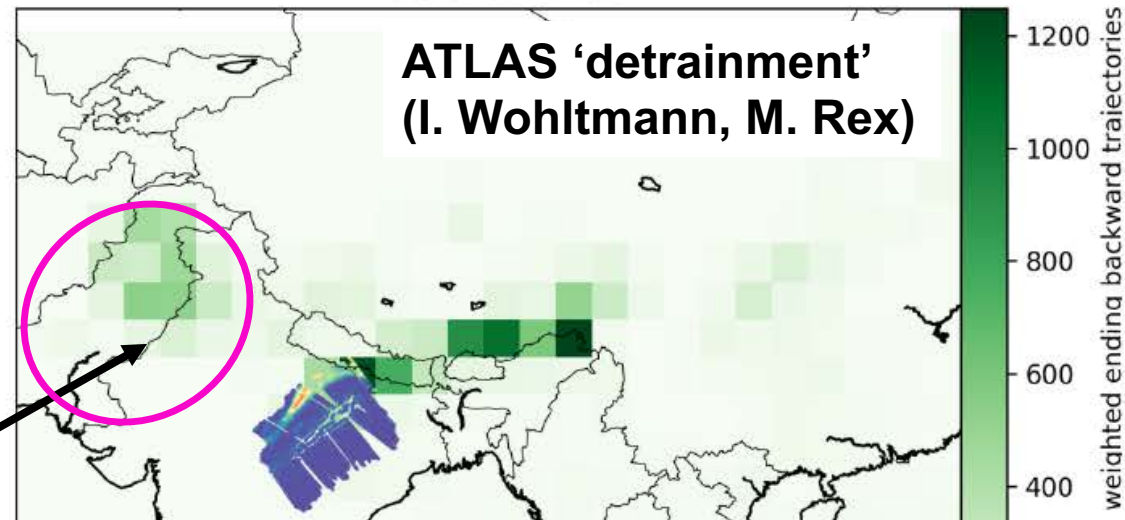
- Different origin of the air encountered on the southbound vs. the northbound flight leg at 13 km altitude
- Same origin at 17 km



2017-07-31:03



2017-07-31:03



Density of backward trajectories originating at position of convection weighted by the observed NH₃ concentrations

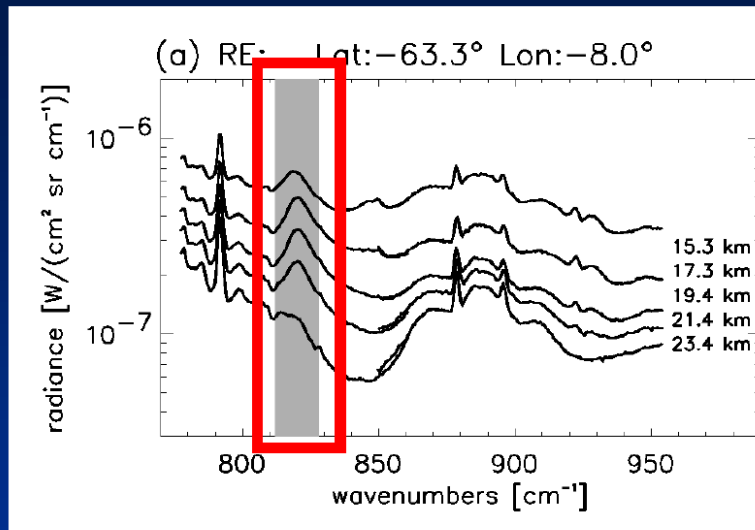
<http://iasi.aeris-data.fr/NH3>
The data produced at Université Libre de Bruxelles (ULB) by Simon Whitburn and Martin Van Damme
Van Damme et al., AMT, 2017

CRISTA and MIPAS observations of Polar Stratospheric Clouds: a peak at 820 cm^{-1}

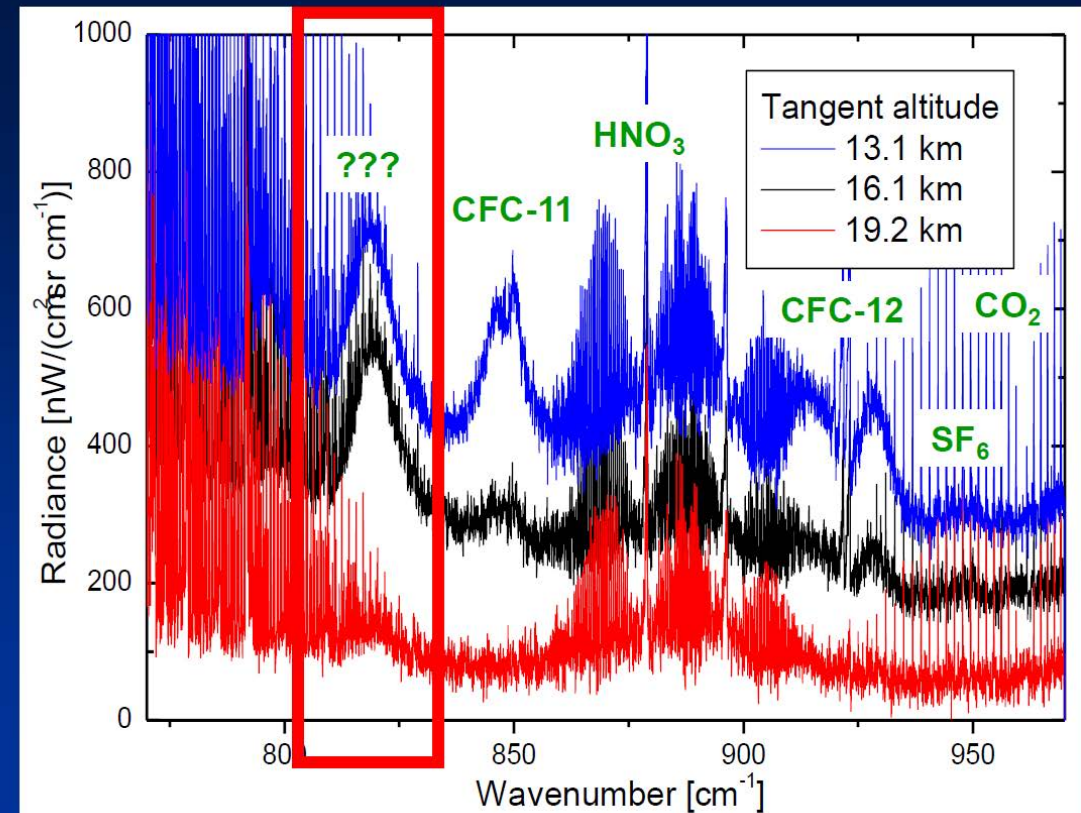
CRISTA PSC observations in August 1997

Spang and Remedios,
2003:

Spectral feature around
 820 cm^{-1} attributed to
NAT indirectly by HNO_3 -
temperature relationship
and occurrence
temperature of the band

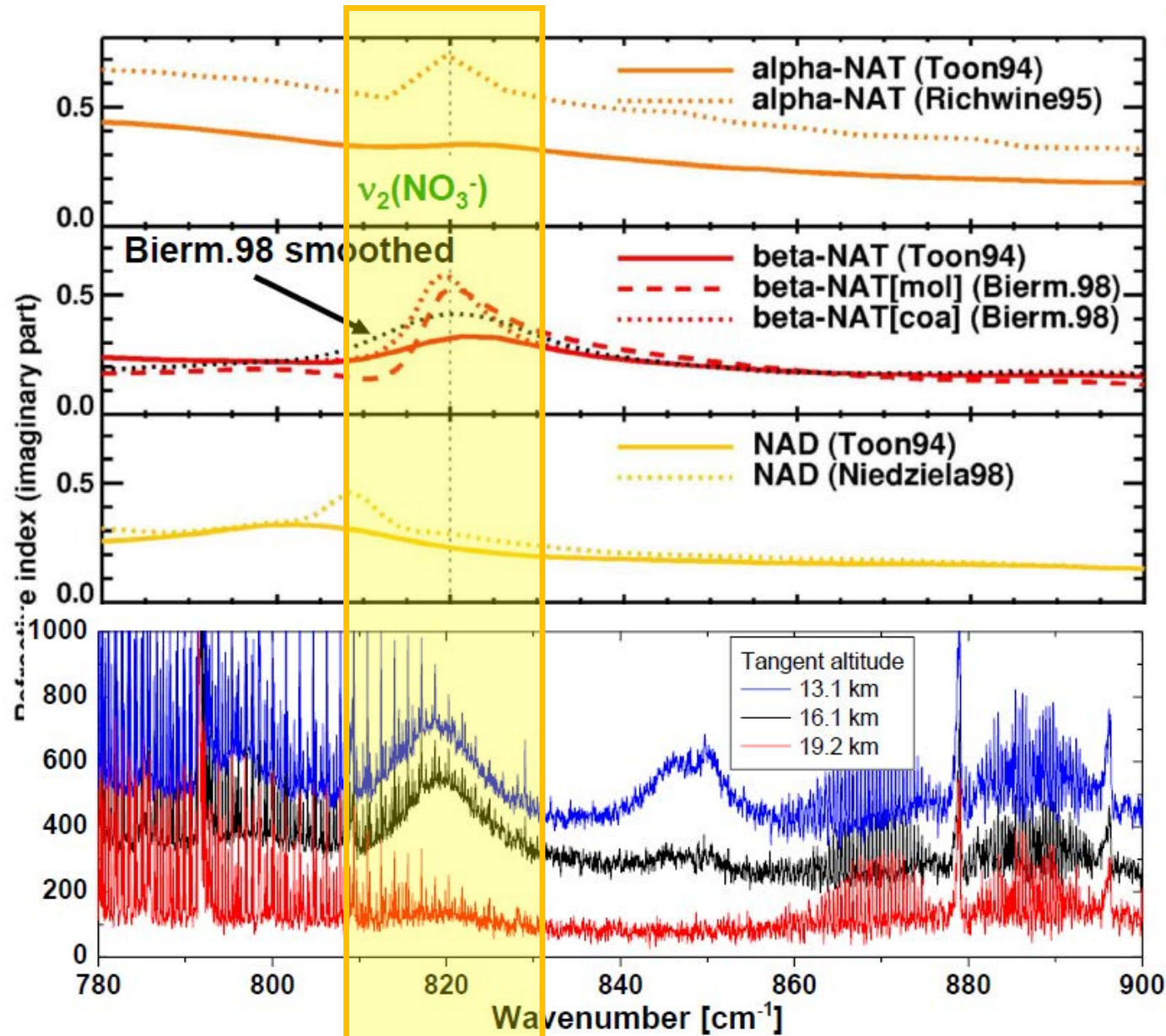


MIPAS PSC observations over McMurdo on 4 Aug 2003



Spectroscopic evidence for NAT, STS, and ice in MIPAS infrared limb emission measurements of polar stratospheric clouds

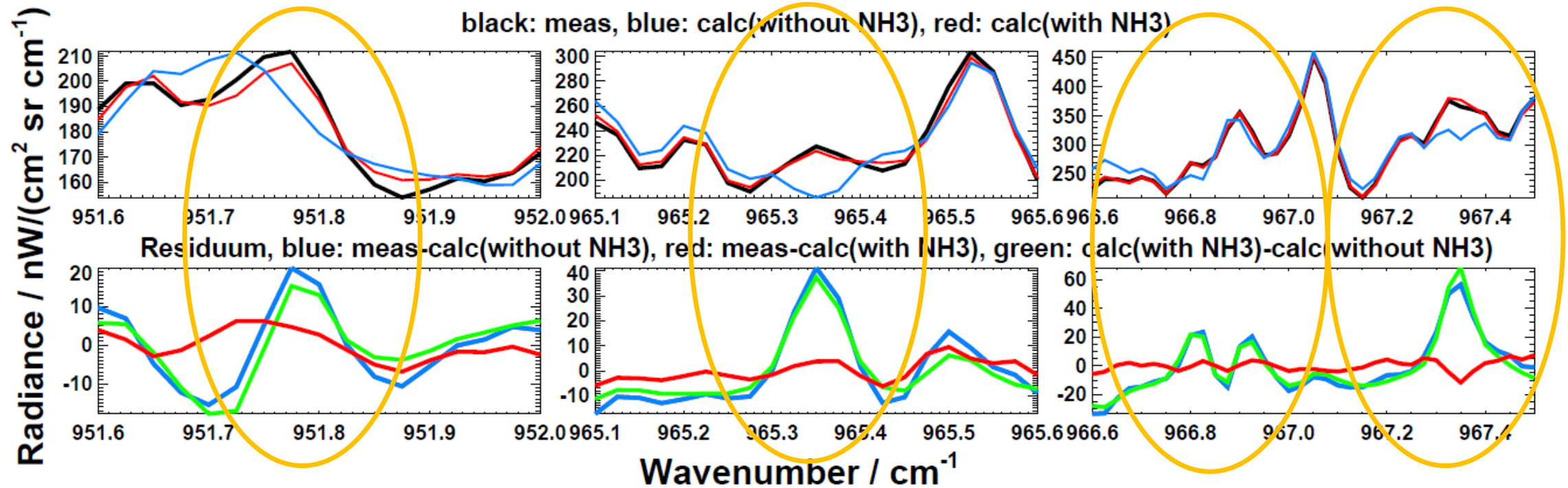
M. Höpfner¹, B. P. Luo², P. Massoli^{3,*}, F. Cairo³, R. Spang⁴, M. Snels³, G. Di Donfrancesco⁵, G. Stiller¹,
T. von Clarmann¹, H. Fischer¹, and U. Biermann^{6,**}



- Infrared signature at 820 cm⁻¹ explained as the $\nu_2(\text{NO}_3^-)$ band of β -NAT

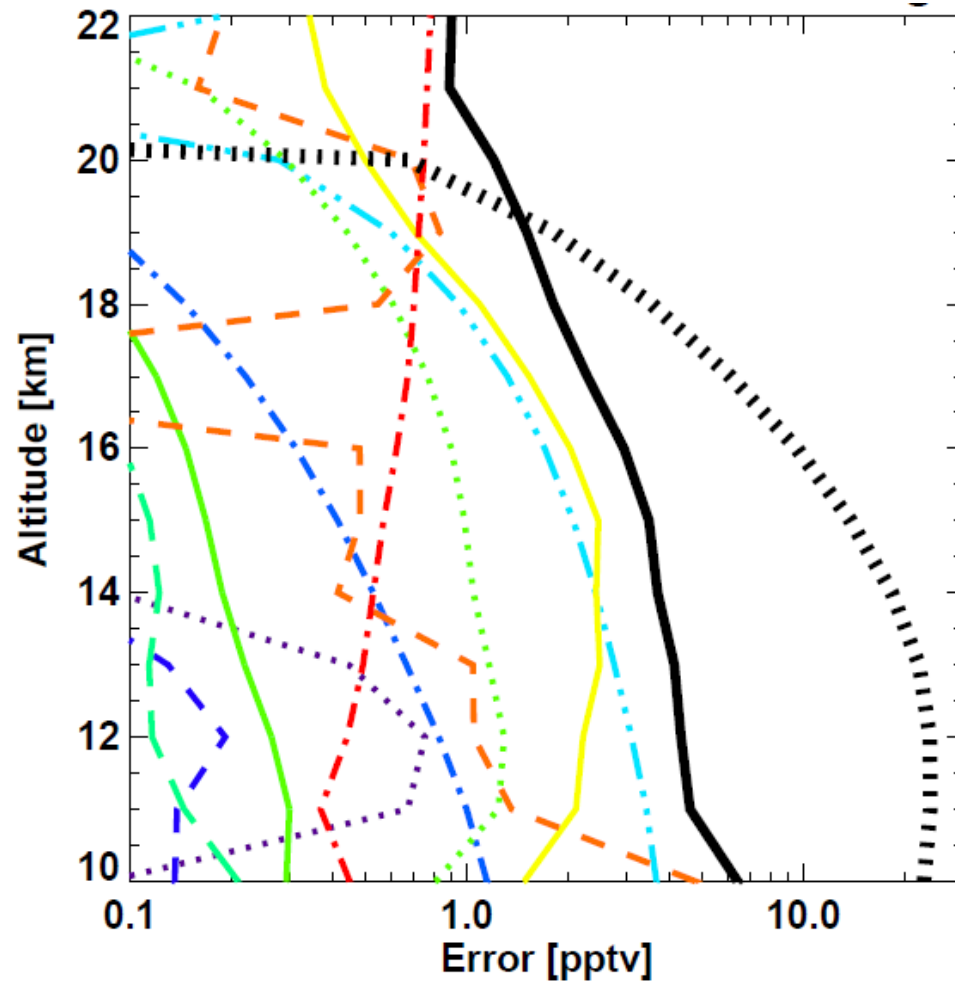
Spectral evidence for NH₃ in the Asian monsoon upper troposphere

Jun/Jul/Aug 2003, latitude range: 30- 40°N, longitude range: 80- 90°E, height: 12.5 km

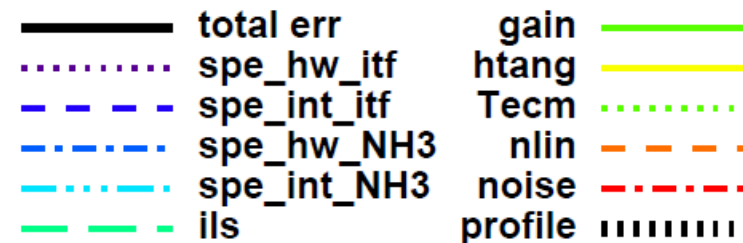


- Clear spectral residuals at position of NH₃ lines
- Strong reduction of spectral residuals when NH₃ is considered

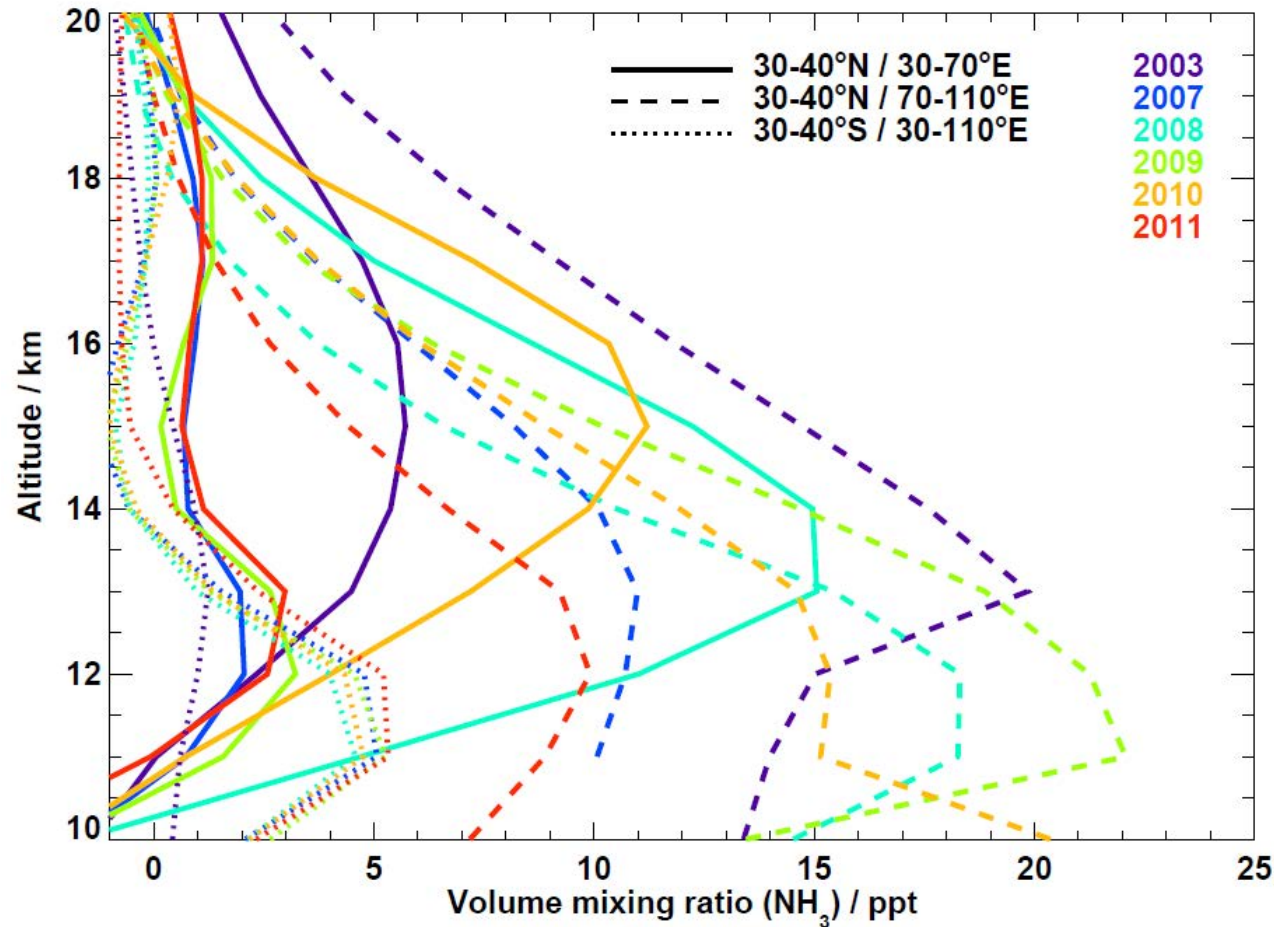
NH₃ retrieval error estimation



- Total retrieval error ~ 5 ppt
- Random noise error ~1 ppt



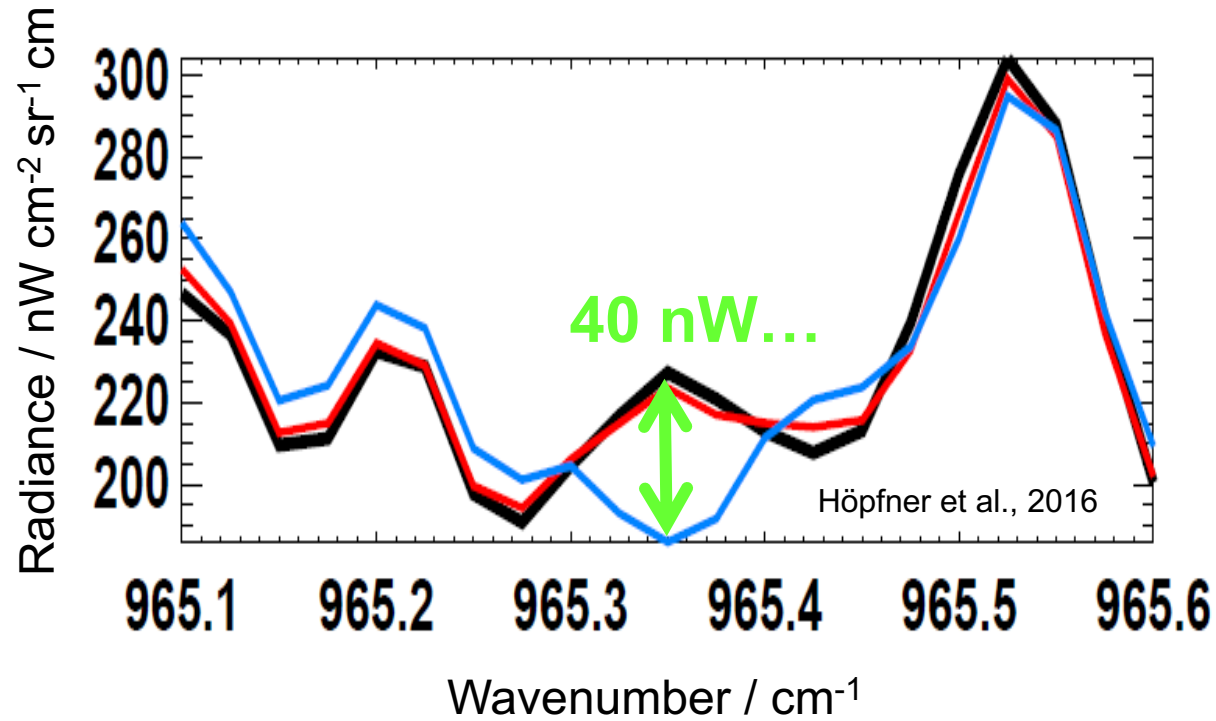
NH₃ profiles within the Asian monsoon



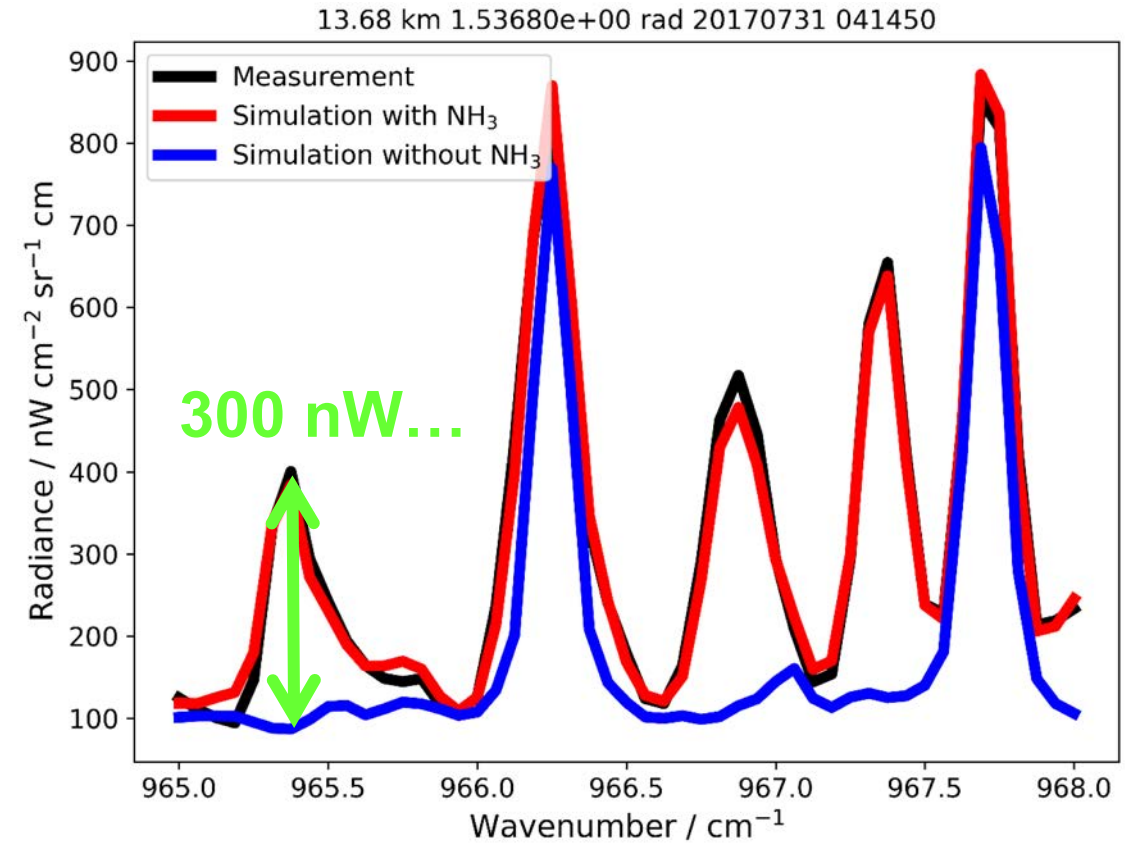
- NH₃ maximum larger and at lower altitudes in the eastern part of the monsoon area
- Much more variable and peaking at higher altitudes in the western part
- Maximum in southern hemisphere indicates detection limit of ~5 ppt

Spectral detection of NH₃

MIPAS/Envisat

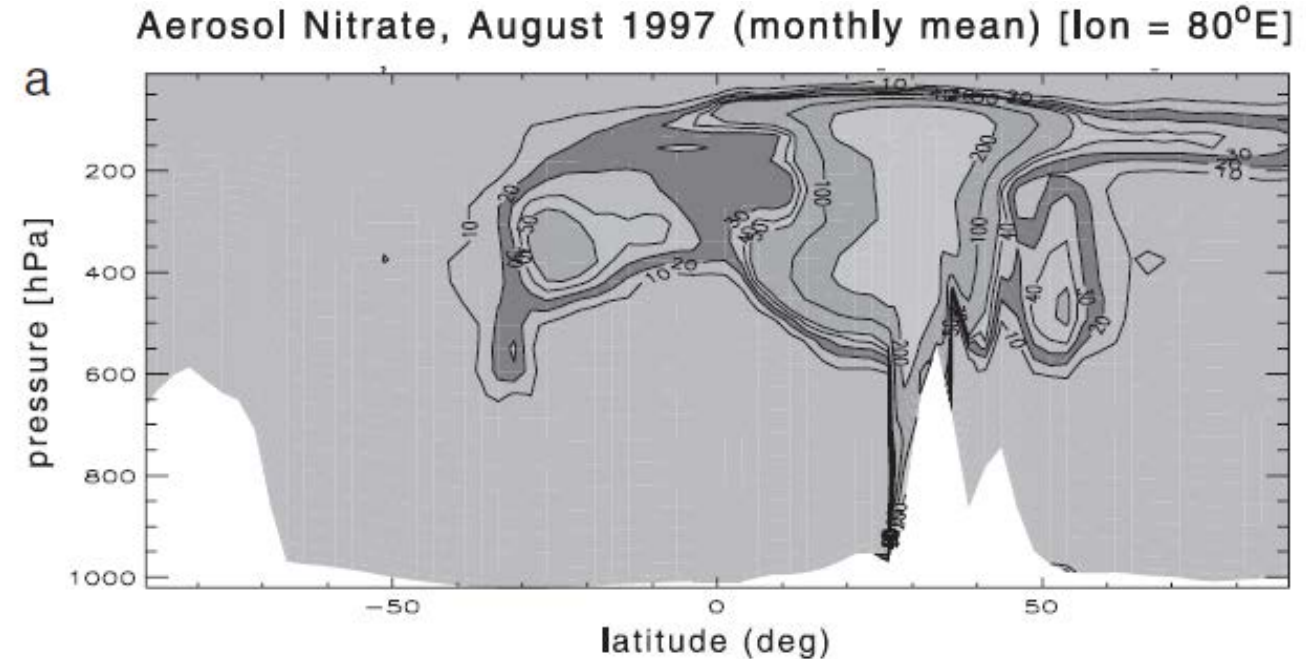


GLORIA/Geophysica



NH₃ and the Asian Tropopause Aerosol Layer

- Model simulates extended plume of ammonium nitrate in the upper troposphere
- Aerosols formed in the UT through neutralization of nitric acid (in the model present in higher amounts than sulfuric acid) by a surplus of NH₃



Metzger et al., JGR, 2002